



## Review

# Development and application of relevance and reliability criteria for water treatment removal efficiencies of chemicals of emerging concern



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## ABSTRACT

With the growth in production and use of chemicals and the fact that many end up in the aquatic environment, there is an increasing need for advanced water treatment technologies that can remove chemicals of emerging concern (CECs) from water. The current lack of a homogenous approach for testing advanced water treatment technologies hampers the interpretation and evaluation of CEC removal efficiency data, and hinders informed decision making by stakeholders with regard to which treatment technology could satisfy their specific needs.

Here a data evaluation framework is proposed to improve the use of current knowledge in the field of advanced water treatment technologies for drinking water and wastewater, consisting of a set of 9 relevance criteria and 51 reliability criteria. The two criteria sets underpin a thorough, unbiased and standardised method to select studies to evaluate and compare CEC removal efficiency of advanced water treatment technologies in a scientifically sound way.

The relevance criteria set was applied to 244 papers on removal efficiency, of which only 20% fulfilled the criteria. The reliability criteria were applied to the remaining papers. In general these criteria were fulfilled with regards to information on the target compound, the water matrix and the treatment process conditions. However, there was a lack of information on data interpretation and statistics.

In conclusion, a minority of the evaluated papers are suited for comparison across techniques, compounds and water matrixes. There is a clear need for more uniform reporting of water treatment studies for CEC removal. In the future this will benefit the selection of appropriate technologies.

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## 1. Introduction

Chemicals are continuously produced for various beneficial purposes, such as protecting crops, conserving food or treatment of diseases. Over 348,000 chemicals are currently registered and regulated via national and international authorities (CHEMLIST). New chemicals enter the market constantly and the global volume of production of chemicals is continuously growing (Bernhardt et al., 2017; CEFIC, 2016; UNEP, 2013). Many of these chemicals and their transformation products enter the aquatic environment during their life cycle (Schwarzenbach et al., 2006). Chemicals of emerging concern (CECs) in the water cycle have been the focus of research for more than 30 years. The main focus has been on assessing their toxicity (Bruce et al., 2010; Schriks et al., 2010), identifying their fate in the environment (Mamy et al., 2015), documenting their occurrence (Loos et al. 2009, 2010, 2013) and minimising their release into the environment together with optimising removal options (Rivera-Utrilla et al., 2013; van Wezel et al., 2017).

To minimise concentrations and thus adverse effects, removal efficiencies of various advanced drinking water and wastewater treatment technologies have been the focus of research (Lee and von Gunten, 2010; Luo et al., 2014; Rivera-Utrilla et al., 2013; Verlicchi et al., 2012; Yang et al., 2017). Advanced water treatment technologies are based on sorption, oxidation and size exclusion principles. The experimental settings in studies on the efficiency of these technologies are not homogeneous. Technologies can be tested at lab-, pilot- or full scale, with different compounds, and under different conditions. Different water matrices can be tested such as demineralised water, real or standardized surface water, ground water, drinking water and wastewater. The target compounds can be spiked as single compound, in mixtures with varying concentrations, or environmental samples can be used. There can be variations in the process conditions of the treatment, e.g. dose, contact time or flux. Finally there are variations in how experimental set-ups and results are expressed; in  $\text{mJ}/\text{cm}^2$  or  $\text{W}/\text{m}^2$  in case of UV oxidation, with Freundlich isotherms or removal percentage in the case of granular activated carbon (GAC), and many others.

These variations, and the resulting ambiguity, obstruct the interpretation and evaluation of data concerning the removal efficiency of CECs of specific treatment technologies. Stakeholders within the urban water cycle have sufficient information on sources, occurrence and risks of CECs and on potential mitigation options, but the relevance and reliability of the information is often unknown (Fischer et al., 2017). As a consequence a framework for evaluation of scientific and technical information when evaluating removal efficiency studies would be helpful, including criteria for

relevance and reliability specified for the technologies. Examples of such data evaluation frameworks from the field of (eco) toxicology are available and well-used, e.g. to identify studies for the derivation of environmental quality standards in a scientifically sound way (Ågerstrand et al., 2011b; Klimisch et al., 1997; Moermond et al., 2016; Roth and Ciffroy, 2016).

The aim of this study is to develop a novel evaluation framework that consists of two criteria sets that can be used to evaluate the i) relevance and ii) reliability of CEC removal efficiency studies for advanced drinking and wastewater treatment technologies. Commonly used advanced drinking and wastewater treatment technologies are activated carbon (granulated or powdered activated carbon, GAC or PAC), the use of ozone ( $\text{O}_3$ ) and UV with or without  $\text{H}_2\text{O}_2$ , and finally nanofiltration (NF) and reverse osmosis (RO) (Luo et al., 2014; van Wezel et al., 2017). Therefore, the criteria developed focus on those technologies. The developed criteria sets are subsequently applied to 244 removal efficiency studies.

## 2. Methodology

Firstly, existing data evaluation frameworks were explored, to be used as a starting point for the water treatment technology evaluation framework.

Secondly, we identified parameters influencing removal efficiencies of the selected treatment technologies, such as CEC characteristics, characteristics of the water matrix involved and the treatment process settings and characteristics. These impact parameters may differ per treatment technology.

An initial literature review was carried out to gain insight in the selected treatment technologies, using Scopus. Papers describing the water treatment techniques GAC/PAC, Ozone/UV  $\pm$   $\text{H}_2\text{O}_2$  and NF/RO membranes were selected based on their title. The full paper was retrieved based on the abstract, and priority was given to recent reviews. Preliminary impact parameter lists were created and subsequently discussed with five experts from universities, knowledge institutes and water utilities, covering all selected technologies. We used semi-structured face-to-face interviews, and e-mailed the preliminary impact parameter lists ahead of the interview, to give the possibility of addressing additional parameters. The interviews were recorded, and analysed. Based on this, further extensive additional literature research on each individual treatment technique was carried out, to underpin the selected impact parameters, and make sure none were left out. A revised impact parameter list was discussed in a workshop within the EU FP7 SOLUTIONS project. Based on this the final impact parameter list was set.

The selected data evaluation framework was then used to develop a novel framework suited to evaluate the CEC removal

efficiency studies for advanced water treatment technologies.

Finally we used the developed framework to evaluate the relevance of 244 removal efficiency papers. These papers were found using the Scopus database by searching on removal of CECs and the chosen treatment technologies, and in the reference lists of the retained articles. The 244 papers evaluated for relevance covered 54 journals, with impact factors from 0.7 to 11.6 and included various research fields such as radiation physics, environmental pollution and chemical engineering (Supplementary Information 1). Papers fulfilling the relevance criteria were evaluated with regards to the reliability criteria.

### 3. Existing data evaluation frameworks

Many frameworks for evaluation of scientific and technical information exist (ECHA, 2011; EPA, 2012; OECD, 1998; USEPA, 2003). Most are developed and applied for the implementation of experimental data in regulatory frameworks or decision making processes. In one of the first generic frameworks for evaluation of scientific and technical information five key factors are highlighted (USEPA, 2003):

1. Soundness - The extent to which the scientific and technical procedures and methods employed to generate the information are reasonable for, and consistent with, the intended application.
2. Applicability and utility - The extent to which the information is relevant for the intended use.
3. Clarity and completeness - The degree of clarity and completeness with which the data, assumptions, methods, quality assurance and analyses employed to generate the information are documented.
4. Uncertainty and variability - The extent to which the quantitative and qualitative variability and uncertainty in the information or in the procedures, measures, methods or models are evaluated and characterized.
5. Evaluation and review - The extent of independent verification, validation and peer review of the information or of the procedures, measures, methods or models.

Besides the generic evaluation frameworks, several scientific fields have their own specific frameworks tailored to their needs. Within the field of (eco)toxicology, thorough and scientifically sound assessment of scientific data is necessary in relation to the hazard and risk evaluation of chemicals (ECHA, 2011; Klimisch et al., 1997; Moermond et al., 2016; Roth and Ciffroy, 2016). In view of the analogy of having to consider the complexity of CECs in various water matrices, these well-developed data evaluation frameworks were considered a useful starting point to develop a data evaluation criteria set for the CEC removal efficiencies.

The first and most used toxicological evaluation framework is the Klimisch approach, focussing on:

1. Reliability - Evaluating the inherent quality of a test report or publication relating to preferably standardized methodologies, and the description of experimental procedures and results to give evidence of the clarity and plausibility of the findings.
2. Relevance - Covering the extent to which data and/or tests are appropriate for a particular intended use of the data, i.e. in the case toxicological evaluation hazard identification or risk characterization.
3. Adequacy - Defining the usefulness of data for the intended (risk assessment) purposes. When there is more than one set of data for each effect, the greatest weight is attached to the most reliable and relevant data set.

The Klimisch approach is used in the REACH legislation for information requirements and chemical safety assessment (ECHA, 2011). The Klimisch framework has been criticised for being too reliant on expert judgement, as it provides few criteria for the reliability evaluation and only mentions relevance with very little guidance on how to evaluate this (Ågerstrand et al., 2011a; Kase et al., 2016; Moermond et al., 2016). Therefore, a more detailed framework for evaluation was developed and tested within the CRED (Criteria for Reporting and Evaluation ecotoxicity Data) project (Kase et al., 2016; Moermond et al., 2016). This framework consists of two criteria sets addressing relevance and reliability. In this framework adequacy is addressed under the heading relevance. The relevance criteria set contains 13 relevance criteria, under the headings general, biological relevance and exposure relevance. The reliability criteria set entails 20 quite specific criteria under the headings general information, test set-up, test compound, test organism, exposure conditions and finally statistical design and biological response. These criteria sets are further elaborated point by point as to why it is important and how it should be verified (Moermond et al., 2016). This elaboration is followed by a recommendation of 50 points which should be included in a study to be able to evaluate it properly.

The existing (eco)toxicology frameworks have recently been evaluated by Roth and Ciffroy (2016). As relevance and reliability were seen as important evaluation points, here we considered only frameworks clearly separating and evaluating these criteria, i.e. Ågerstrand, AMORE and CRED (Ågerstrand et al., 2011b; Isigonis et al., 2015; Moermond et al., 2016). The AMORE framework was not selected as it is a computer based decision support system very specifically tailored to the evaluation of ecotoxicity tests and not easily adaptable to other purposes (Ågerstrand et al., 2011b; Isigonis et al., 2015; Moermond et al., 2016; Roth and Ciffroy, 2016). The CRED framework gives a detailed description of what needs to be assessed for the evaluation of relevance and reliability, summarized in two tables. These were used as a starting point to develop a relevance and a reliability criteria set for the evaluation of CEC removal efficiency studies for water treatment technologies.

## 4. Results

### 4.1. Impact parameters for water treatment efficiency assessment

The impact parameters which influence CEC removal efficiencies of advanced water treatment technologies concern a) CEC characteristics, b) water matrix characteristics and c) treatment process conditions. These are specified per treatment technology in Table 1. Parameters in bold are unique to each study, other parameters are relevant but can be retrieved elsewhere such as via scientific literature, and databases such as Episuite, Chemspider and OECD toolbox.

### 4.2. Relevance and reliability criteria for water treatment efficiency assessment

#### 4.2.1. Assessing the relevance of a water treatment study

For relevance we used the definitions of Klimisch et al. (1997) and ECETOC (2009) "Relevance broadly refers to the extent to which **data** and **tests** are **appropriate (fit-for-purpose)** for their intended use, it is a context-dependent quality criterion that is neither intrinsic to a given study per se, nor a function of the information available about that study". The relevance criteria, **appropriateness of data** and **tests** (see Table 2), are used to determine whether a paper is of interest for the specific purpose; if positive the reliability should be assessed.

**Table 1**

Impact parameters per treatment technology. Parameters in bold are essential to report, other parameters can be retrieved elsewhere.

Treatment technology	CEC characteristics	Water matrix characteristics	Treatment process conditions	References
Granular activated carbon	Molecular charge/pKa Log $K_{ow}$ /Log $D_{ow}$	pH <b>NOM<sup>a</sup> concentration (and composition)</b> <b>Temperature</b>	<b>Surface area/grain size</b> <b>Pore volume</b>	De Ridder et al., 2011; Jeirani et al., 2017; Mailler et al., 2016; Nam et al., 2014; Rossner et al., 2009; Verlicchi et al., 2010
	Molecular weight/size Functional groups (H-bonds, aromaticity etc.) <b>Concentration</b>		Surface charge <b>Biological activity</b>  <b>Contact time/EBCT<sup>b</sup></b> <b>Column length</b> <b>Flow through</b> <b>Backflush routine</b> <b>Scale of testing (bench, pilot, full)</b> <b>Prior carbon use (if any)</b>	
Powdered activated carbon	Molecular charge/pKa Log $K_{ow}$ /Log $D_{ow}$	pH <b>NOM<sup>a</sup> concentration (and composition)</b> <b>Temperature</b>	<b>Surface area</b> <b>Pore volume</b>	Lee et al., 2013; von Sonntag and von Gunten, 2012; Zhang et al., 2012
	Molecular weight/size Functional groups (H-bonds, aromaticity etc.) <b>Concentration</b>		<b>Contact time</b> <b>Concentration</b>  Surface charge <b>Scale of testing (bench, pilot, full)</b> <b>Dosage of O<sub>3</sub> (and H<sub>2</sub>O<sub>2</sub>)</b> <b>Reactor design (mixing regime)</b> <b>Contact time</b> <b>Scale of testing (bench, pilot, full)</b>	
Ozone (+H <sub>2</sub> O <sub>2</sub> )	Reactivity <b>Concentration</b>	pH <b>NOM<sup>a</sup> concentration</b>  <b>Nitrite/nitrate</b> <b>Bromide/bromate</b>	<b>Wavelength (lamp type)</b> <b>Irradiation time</b> <b>UV dose and H<sub>2</sub>O<sub>2</sub> dosage</b> <b>Reactor design</b> <b>Scale of testing (bench, pilot, full)</b> <b>Membrane area</b> <b>Membrane charge/Zeta potential</b> <b>Fouling</b>	Lester et al., 2008; Liu and Liu, 2004; Pereira et al., 2007; Real et al., 2009; Yang et al., 2014
UV (+H <sub>2</sub> O <sub>2</sub> )	Reactivity <b>Concentration</b>	pH <b>NOM<sup>a</sup> concentration</b> <b>Nitrite/nitrate</b> <b>Temperature</b> <b>Turbidity</b>	<b>Trans Membrane Pressure</b> <b>Cross-flow velocity (only NF)</b> <b>Permeate flux</b> <b>Recovery</b> <b>Salt rejection data (Molecular weight cut-off)</b> <b>Scale of testing (bench, pilot, full)</b>	Bellona et al., 2004; Hajibabania et al., 2011b; Verlicchi et al., 2007b; Yoon et al., 2006
NF/RO membranes	Molecular charge/pKa Molecular weight (size)	<b>Ionic strength</b> pH		
	Functional groups	<b>NOM<sup>a</sup> concentration (and composition)</b> <b>Temperature</b> <b>Turbidity</b>		
	Log $K_{ow}$ /Log $D_{ow}$ <b>Concentration</b>			

<sup>a</sup>NOM: Natural Organic Matter, <sup>b</sup>EBCT: Empty Bed Contact Time.

#### 4.2.1.1. Explanation of the relevance criteria (criteria numbers from Table 2). **Relevance of data**

##### 1 Is the scope of the tests appropriate for the evaluation?

The requirements for evaluating CEC removal efficiencies is that the technology is commercially available and full-scale applied in the water sector. Only then can it be used by stakeholders to make informed decisions about relevant treatment technologies. This at current implies that CEC removal studies on activated carbon (GAC and PAC), O<sub>3</sub> ( $\pm$ H<sub>2</sub>O<sub>2</sub>), UV ( $\pm$ H<sub>2</sub>O<sub>2</sub>) or NF and RO membranes are considered appropriate. Evidently, in the future this list of commercially available and full-scale applied techniques can be expanded.

##### 2 Are the data reported appropriate for the evaluation?

The purpose is to compare the removal efficiency of one treatment technology with another. This means that if a study reports removal efficiencies in such a way that these cannot be compared across technologies, e.g. break-through curves for granular activated carbon, it is not considered relevant for this purpose. The

results should ideally be reported as removal percentage, log units of removal, or influent/effluent concentrations which can be used to calculate the removal efficiency. Studies presenting the removal percentages only in graph form were disregarded as exact removal percentages cannot be retrieved from graphs. Many studies use the connotation > and < before the removal percentage as they cannot be more specific due to the limit of quantification/limit of detection (LOD/LOQ), this severely hampers the possibility of comparing the study to other studies. In case > or < have been used, only studies reporting removal percentages >99% or <1% have been included, as these results are considered equal to 99% or 1% removal. In case of negative removal efficiencies these may reflect < LOQ in influent matrices, fluctuating concentrations, or transformation processes during treatment, and can be relevant to include. However, studies to find the conditions at which a certain percentage of a compound is removed, i.e. where the removal percentage is fixed, are discarded as this is a different way of studying removal.

#### **Relevance of tests**

**Table 2**

Criteria for assessing the relevance of efficiency studies for advanced water treatment technologies.

Criterion	#
<b>Relevance of data</b>	
Is the scope of the tests appropriate for the evaluation?	1
Are the data reported appropriate for the evaluation?	2
<b>Relevance of tests</b>	
Relevance of target compounds	
In case of a formulation or mixture is the compound tested representative and appropriate for the compounds being assessed?	3
Are the properties of the compounds chosen appropriate for the purpose of the tests?	4
Is the applied compound concentration appropriate for the purpose of the tests?	5
Relevance of water matrix	
Is an appropriate type of water chosen for the purpose of the tests?	6
Are the properties of the water matrix chosen appropriate for the purpose of the tests?	7
Relevance of treatment technology	
Are the properties of the treatment technology chosen appropriate for the purpose of the tests?	8
Is the scale of the experiment appropriate for the purpose of the tests?	9

### 3-5 Relevance of target compound(s)

When testing a large number of compounds these should ideally represent a broad range of physicochemical properties. The focus can also be on one compound or a group of compounds that are known to be problematic to remove with other techniques than advanced water treatment techniques, such as persistent mobile organic compounds (PMOCs) (Reemtsma et al., 2016). In case of compounds that degrade easily due to other processes than the treatment studied, it should be verified that the compound is actually removed by the tested treatment technology. This emphasizes the importance of controls. Ideally tested concentrations of the CECs should be representative for environmental levels, but this can be difficult to achieve with compounds with a high LOQ.

### 6-7 Relevance of water matrix

It is important to ensure that the selected water matrix corresponds with the purpose of the study. Especially wastewater has many properties that alter removal efficiencies compared to cleaner matrices (Luo et al., 2014). As an example the often higher wastewater content of DOC/NOM influences the efficiency of O<sub>3</sub> or UV compared to surface water with generally lower DOC/NOM content.

### 8-9 Relevance of treatment technology

Experimental conditions need to be appropriate for the purpose of the tests. It should be considered how the tested conditions relate to full-scale operational ranges (e.g. dosing, exposure time) and whether they are realistic. For example, the maximum dosage of O<sub>3</sub> is in many cases legally restricted because of the formation of bromate that is highly toxic (von Gunten, 2003b).

## 4.2.2. Assessing the reliability of a water treatment study

When assessing the reliability of a water treatment study for data evaluation, we used the definition by Roth and Ciffroy (2016): “The reliability of a study relates inter alia to the robustness and validity of the **method** used, the completeness and detail of **reporting**, the clarity and plausibility of the findings to ensure their **reproducibility**, but also to the **uncertainty** of the knowledge base”. With regard to reliability, we divide the criteria into the highlighted components from the above definition: **method, reporting information, reproducibility and uncertainty** (see Table 3).

### 4.2.2.1. Explanation of the reliability criteria (criteria numbers from Table 3). Method

1-2 Is a guideline or modified guideline used for any part of the experiment? Is the test performed under GLP conditions?

The International Organization for Standardization (ISO) and the Organisation for Economic Co-operation and Development (OECD) are continuously producing guidelines to amongst others

standardize various chemical tests (ISO, 2017; OECD, 2017). At the moment no guidelines for testing removal efficiencies for the selected treatment techniques are available. Nevertheless, ISO and OECD have provided guidelines on how to measure CECs and how to take water samples. The American Society for Testing and Materials (ASTM) published several standards related to water treatment technologies on their website, amongst others standards on how to predict adsorption capacity of activated carbon (AC) and some standard procedures for membrane testing e.g. salt rejection (ASTM, 2014a; b; c). These guidelines are relevant for certain studies on CEC removal efficiency. Besides these guidelines, it is also relevant to know if the experiments are performed under good laboratory practices (GLP) conditions.

### 3 Are validity criteria fulfilled (controls)?

This to verify that only the tested treatment technology is removing the CECs from the water matrix and no other processes such as natural degradation or evaporation takes place.

### Reporting information

To have all the information necessary to make a thorough evaluation of the limitations and exact conditions of the CEC removal efficiency study the following information (criteria 4-47) should be reported either in the study itself or in the supplementary information.

### Target compound

#### 4-6 Identification

Many of the below mentioned impact parameters can be found in databases and literature (e.g. Episuite, Chemspider, OECD toolbox etc.) if the compound name and/or CAS number is known.

#### 7 Molecular charge/pKa (AC, Ozone, NF/RO membranes)

The pKa of the compound combined with the pH of the water matrix can be used to determine the charge of the compound if this is not given in the study (de Ridder et al., 2010; Moreno-Castilla, 2004). A charged molecule will be either attracted to or repelled by the charge of the AC surface (De Ridder et al., 2011; de Ridder et al., 2010; Kovalova et al., 2013; Mailler et al., 2014; Margot et al., 2013; Moreno-Castilla, 2004). For the same reason the charge of a compound also has an influence on the NF/RO rejection, usually negatively charged CECs are better rejected than positively charged CECs (Bellona et al., 2004; Verliefde et al., 2008; Yoon et al., 2007). Also for treatment with ozone the charge is often relevant, especially for nitrogen containing compounds as often only deprotonated compounds have enough electron density, see also 10 Functional groups/reactivity, to be attacked by ozone (Borowska et al., 2016).

#### 8 Log K<sub>ow</sub>/Log D<sub>ow</sub> (AC, NF/RO membranes)

The hydrophobicity of a compound affects the efficiency of a treatment technique (Hu et al., 1998; Kovalova et al., 2013; Mailler

**Table 3**

Criteria for assessing the reliability of removal efficiency studies for advanced water treatment technologies, specified per treatment technology. When no technology is specified the criterion applies to all technologies.

Criterion	#
<b>Method</b>	
Is a guideline (e.g., OECD <sup>a</sup> /ISO <sup>b</sup> /ASTM <sup>c</sup> ) or modified guideline used for any part of the experiment?	1
Is the test performed under Good Laboratory Practices (GLP) conditions?	2
Are validity criteria fulfilled (e.g. controls)?	3
<b>Reporting information</b>	
<b>Target compound</b>	
Identification	
Name	4
CAS number or other identifier	5
Form tested (e. g. salt, acid or base)	6
Impact parameters	
Molecular charge/pKa (AC, ozone, NF/RO membranes)	7
Log K <sub>OW</sub> /Log D <sub>ow</sub> (AC, NF/RO membranes)	8
Molecular weight/size (AC, NF/RO membranes)	9
Functional groups/reactivity	10
Supplier, purity of target compound	11
<b>Water matrix</b>	
Matrix being wastewater, wastewater effluent, surface water, ground water, drinking water, demineralized water or synthetic wastewater	12
Impact parameters	
pH	13
NOM concentration and possibly composition	14
Temperature	15
Nitrite/nitrate concentration (ozone, UV)	16
Bromide/bromate concentration (ozone)	17
Turbidity (NF/RO membranes)	18
UV transmittance (UV)	19
<b>Treatment process conditions</b>	
Identification	
Type of technology used	20
Impact parameters	
Surface or membrane area (AC, NF/RO membranes)	21
Pore volume and pore size distribution (AC)	22
Bed volume (GAC)	23
Flow through (GAC)	24
Preloading (GAC)	25
Surface charge (AC, NF/RO membranes)	26
Concentration/dosage or intensity (PAC, ozone, UV)	27
Contact, exposure or irradiation time (AC, ozone, UV)	28
Biological activity (GAC)	29
Technology application (diagram)	30
Wavelength (UV)	31
Concentration of reagent H <sub>2</sub> O <sub>2</sub> (ozone, UV)	32
Fouling (NF/RO membranes)	33
Trans membrane pressure (NF/RO membranes)	34
Permeate flux (NF/RO membranes)	35
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<sup>a</sup> OECD: The Organisation for Economic Co-operation and Development.

<sup>b</sup> ISO: International Organization for Standardization.

<sup>c</sup> ASTM: The American Society for Testing and Materials.

et al., 2014; Westerhoff et al., 2005). The hydrophobicity is often expressed as the octanol-water partitioning coefficient Log K<sub>ow</sub>, or the pH corrected log D<sub>ow</sub> (De Ridder et al., 2011; Kovalova et al., 2013). Log D<sub>ow</sub> values can be calculated on the basis of the

reported pH of the water matrix, the pK<sub>a</sub> and the log K<sub>ow</sub>, so it is not essential to report. For NF/RO membranes more hydrophobic CECs can be adsorbed by the membrane and are initially well removed, but removal may decrease later due to saturation of the membrane

(Hu et al., 2007; Moons and Van der Bruggen, 2006; Verliefde et al., 2007a).

#### 9 Molecular weight/size (AC, NF/RO membranes)

Large compounds tend to be less easily removed with AC (De Ridder et al., 2011; Mailler et al., 2014). The size of a compound, expressed in molecular weight, is believed to be one of the main factors in rejection of CECs by NF/RO membranes (Bellona et al., 2004; Yoon et al., 2007). Besides the molecular weight also the molecular size, width, length, (effective) diameter and the molecular volume are used in studies as a measure of the size of the CEC (Braeken et al., 2005; Comerton et al. 2008, 2009; Jung et al., 2005; Kiso et al., 2001; Sadmani et al., 2014; Yangali-Quintanilla et al., 2009).

#### 10 Functional groups/reactivity (All)

Relevant functional groups can be identified on the basis of the compound structure, so in principle they do not have to be reported although it is helpful if the author includes information on this topic. Different functional groups are relevant for different techniques. With regards to AC especially groups allowing H-bonds or  $\pi$ - $\pi$  binding are relevant (de Ridder et al., 2010; Mailler et al., 2014). For NF/RO membranes functional groups of CECs may play a role in whether the compound is rejected or not, this is, however, mostly related to the resulting size and/or charge of the compound (Bellona et al., 2004; Košutić and Kunst, 2002; Ozaki and Li, 2002). With regards to O<sub>3</sub> and UV, it is the reactivity of a compound that is important. This is related to the electron density of the compound, which can again be related to the functional groups. Functional groups with a high electron density will increase the reactivity of a compound (von Sonntag and von Gunten, 2012). For treatment with ozone a broad range of functional groups such as nitro groups, amides, primary amines, thioethers and olefins are known to enhance the removal of CECs (Kovalova et al., 2013; Nakada et al., 2007; Snyder et al., 2007a; von Gunten, 2003a). For UV or O<sub>3</sub> treatment in combination with H<sub>2</sub>O<sub>2</sub> an even broader scale of chemical structures will be attacked as OH radicals react more unspecific with a very wide variety of functional groups (Snyder et al., 2007a).

#### 11 Supplier and purity (All)

Supplier and purity of purchased target compounds should be reported, as this cannot be retrieved elsewhere. This is important with regards to both the analysis and the behaviour of the target compounds.

#### Water matrix

##### 12 Type of water (All)

It is essential to know which type of water was used and whether it has been artificially created or is of environmental origin. This gives an indication of the relevance of the study in relation to full scale treatment and what issues to expect from the water matrix. If the water matrix is artificially created, details on the substituents and preparation procedure are to be reported. In addition, the origin of the composition may be given, as this can be based on literature e.g. greywater (Benami et al., 2016; Pradhan et al., 2019) and wastewater (Wilsenach and Van Loosdrecht, 2004).

##### 13 pH (All)

As previously discussed under 7 *Molecular charge/pKa*, the pH of the water matrix is relevant. In most full scale treatments the pH lies between 7 and 9 (Kovalova et al., 2013; Mailler et al., 2014; Margot et al., 2013), so it should be explicated if the study has been performed at a significantly different pH. Depending on the pKa a CEC might express higher or lower reaction rates with UV or ozone at a specific pH due to whether it will be present in a deprotonated or protonated state, as mentioned in 7 *Molecular charge/pKa* (Avisar et al., 2010; Borowska et al., 2016; Real et al., 2009). Furthermore, at a high pH, ozone decomposition to OH radicals increases, and this likely will change the CEC removal (Real et al., 2009). The pH can

also have an influence on the surface charge of both NF/RO membranes and AC, as well as impact fouling of the membrane (Bellona et al., 2004; Braghetta et al., 1997; Deshmukh and Childress, 2001; Hagemeyer and Gimbel, 1998).

##### 14 NOM concentration (and composition) (All)

For all treatments the presence of NOM can reduce the efficiency of the treatment in which both the concentration and composition are relevant parameters (Boehler et al., 2012; De Ridder et al., 2011; Loos et al., 2013; Mailler et al., 2016; von Gunten, 2003a). With regards to AC NOM can reduce the removal efficiency of CECs by pore blocking or adsorption competition (Kilduff et al., 1998; Matsui et al., 2002; Newcombe and Drikas, 1997; Pelekani and Snoeyink, 1999). The composition of NOM has proven to be relevant for adsorption competition, where protein-like fluorophores seem to be the most problematic molecular components (Mailler et al., 2016). In advanced oxidation treatment the OH radicals are scavenged by NOM, leading to lower removal efficiencies (Neamțu and Frimmel, 2006; Pereira et al., 2007; Snyder et al., 2007a; von Gunten, 2003a; Yuan et al., 2009; Zwiener and Frimmel, 2000). In the case of UV treatment the presence of NOM can increase the removal efficiency of the treatment as the presence of NOM can initiate the production of OH radicals (Leech et al., 2009; Neamțu and Frimmel, 2006; Pereira et al., 2007; Zhan et al., 2006). On the other hand NOM can also decrease the efficiency of UV treatment as it can adsorb the UV, see 19 *UV transmittance*. As with UV the effect of NOM on NF/RO membrane removal efficiency is ambiguous, depending on the amount of NOM, the type of CECs, and the type and thickness of the fouling layer, and has been researched in many studies (Comerton et al., 2009; Nghiem and Coleman, 2008; Nghiem et al., 2010; Nghiem and Hawkes, 2007; Verliefde et al., 2009a). Usually NOM in wastewater is negatively charged, assisting in repulsion of negatively charged CECs (Mailler et al., 2014). Even though the effect of NOM present in the water matrix may not be clear, it is important to mention the presence, concentration, and when relevant and possible the composition of the NOM, so the influence can be taken into consideration when evaluating the removal efficiency. In the 49 studies evaluated in this paper NOM is expressed as DOC (18 studies), TOC (17 studies) and/or COD (13 studies). To enable comparison across studies it is recommended to include at least DOC. Only 4 studies mentions NOM composition, 2 PAC studies, one NF and one NF/RO study.

##### 15 Temperature (All)

With regards to AC the temperature can influence adsorption kinetics, with a low temperature giving lower removal efficiency (Nam et al., 2014). In removal with ozone, temperature can influence ozone decay. In a study by Real et al. (2009) the rate constants of removal with UV increased with increasing temperature. However, the difference in minimum and maximum temperature in the study was 30 °C, and this difference is unlikely to occur in full-scale applications. With regards to NF/RO membranes a higher temperature will result in more viscous water, which will easier transport (diffuse) through the membrane and therefore result in a higher flux which also influences the CEC removal, see also 34-35 *Transmembrane pressure and permeate flux* (Crittenden and Montgomery Watson, 2012).

##### 16-17 Nitrite/nitrate and bromide/bromate (Ozone, UV)

When applying ozone (especially for wastewater) the nitrite concentration is relevant since nitrite reacts quickly with ozone and reduces the available dose (Hollender et al., 2009; Lee et al., 2013). For both UV and ozone the nitrate concentration (especially for drinking water) is also relevant as this can form nitrite and other by-products which are highly toxic. Furthermore, bromide can form the toxic compound bromate with ozone and consequently, above certain bromide concentrations ozonation cannot be recommended (Lee et al., 2013; von Gunten, 2003b; von Sonntag and von

Gunten, 2012).

#### 18 Turbidity (NF/RO membranes)

For NF/RO membranes the turbidity can give an indication of the pre-treatment needed and whether to expect fouling, see 33 *Fouling* (Kim et al., 2002).

#### 19 UV transmittance (UV)

The UV transmittance of the water matrix is relevant to be able to understand the ability of the UV radiation to reach the CECs. The transmittance is an indication of how much of the UV radiation is adsorbed by dissolved matter in the water matrix, and is related to the NOM and nitrate content of the water matrix (Kruithof and Martijn, 2013). In general, the higher the transmittance the higher the CEC removal (Wols et al., 2015). A low transmittance can be compensated for by using more energy to reach the same UV dose.

#### Treatment process conditions

The parameters listed below are all relevant to report as they are unique to the particular study and cannot be found elsewhere. Some of these parameters will change over time, so it is important to report any development of these parameters over the course of the technology test.

#### 20 Type of technology (All)

The exact type of water treatment technology used should be reported.

#### 21 Surface/membrane area (AC, NF/RO membranes)

The surface area has a significant influence on the removal efficiency of a specific AC, and is thus important to report. In general, a smaller surface has a smaller removal efficiency (Mailler et al., 2016). The surface area can be acquired from the producer, if the commercial name is provided in the study. For NF/RO membranes the membrane area tested is important to understand the scale of the test rather than its influence on the removal capacity (Verliefde et al., 2009b).

#### 22 Pore volume and pore size distribution (AC)

The pore volume and the pore size distribution is relevant in relation to the size of the CEC, and the competition of NOM (Kovalova et al., 2013; Mailler et al., 2016; Moreno-Castilla, 2004). NOM can access mesopores and block these and the access of the CECs to underlying micropores (Moreno-Castilla, 2004).

#### 23-25 Bedvolume, flow through and preloading (GAC)

The bedvolume and flow through of the GAC column influences the removal efficiency of the GAC treatment (Chowdhury, 2013). Bedvolume and flow through indicate the time the CECs have to adsorb to the AC, see also 28 *Contact time*. In case the GAC has been used before it will be preloaded with NOM, this should be reported as it can have a negative influence on the efficiency, however preloaded GAC columns might also exhibit biological degradation which will have a positive effect on the removal efficiency (Bourgin et al., 2018).

#### 26 Surface charge (AC, NF/RO membranes)

The impact of the surface charge has been discussed under 7 *Molecular charge/pKa*. For NF/RO membranes the surface charge is often quantified as zeta potential (Bellona et al., 2004). NF/RO membranes are mostly negatively charged at neutral pH, and thus have a higher affinity for positively, or neutral compounds (Bellona et al., 2004; Taheran et al., 2016).

#### 27 Concentration/dosage/intensity (PAC, ozone, UV)

For PAC, the dosage is the main controlling factor for the removal efficiency, the higher the dose the higher the removal (Boehler et al., 2012; Mailler et al., 2014; Snyder et al., 2007b). Also for ozone, the dosage has a substantial positive influence on the removal efficiency (Margot et al., 2013; Snyder et al., 2007a; Zwiener and Frimmel, 2000). Often the ozone dosage is related to the DOC concentration of the water matrix, as a higher dose can compensate for the limiting effect of the DOC concentration

(Zwiener and Frimmel, 2000). Especially for wastewater the ozone dosage is often given in relation to the DOC concentration as g O<sub>3</sub>/g DOC (Hollender et al., 2009; Kovalova et al., 2013). Also for UV in general the higher the dose the higher the removal efficiency, however, this is only valid for CECs that are already susceptible to degradation with UV light (Kovalova et al., 2013; Wols and Hofman-Caris, 2012; Wols et al., 2013). In some of the studies evaluated the UV intensity is given in einstein s<sup>-1</sup>, which is not a SI unit and cannot be compared to the more commonly used mj/cm<sup>2</sup>.

#### 28 Contact, exposure or irradiation time (AC, ozone, UV)

The contact time can have a great influence on the removal efficiency of AC (Mailler et al., 2016; Snyder et al., 2007b). However for compounds with very fast adsorption kinetics this impact factor will be less relevant (Mailler et al., 2016). In UV studies the irradiation time is included in the calculation of the dose, so it is less relevant to state. However, when using advanced oxidation, the exposure time, during which the OH radicals have time to form and react with the CECs, is relevant.

#### 29 Biological activity (GAC)

Usually it is known if there is a possibility of biological activity in the GAC experiments. If this is the case it needs to be stated as the presence of microorganisms in the GAC filter can enhance removal of biodegradable CECs (Bourgin et al., 2018; Magic-Knezev and van der Kooij, 2004; Rattier et al., 2014). Biological activity depends on the time the GAC have been in use. When using fresh GAC or GAC directly after regeneration there will be no biological activity in the beginning, but it will develop and increase in time during use.

#### 30 Technology application (All)

The way the technology is applied is also relevant to report as for example the efficiency of ozone depends a lot on the mixing regime (Dodd et al., 2008; Zucker et al., 2016). It can be difficult to describe an experimental set-up, so in this case the criterion is considered fulfilled if a schematic of the set-up has been included.

#### 31 Wavelength (UV)

The wavelength or the source of the UV light (which type of lamp) is relevant, especially in the case of treatment without H<sub>2</sub>O<sub>2</sub> (Yang et al., 2014). As mentioned under 10 *Functional groups/reactivity*, this can have influence on whether the CEC will be able to absorb the radiation, and thus degraded by photolysis (Chen et al., 2007; Rosenfeldt and Linden, 2004; Yang et al., 2014). Furthermore a broader UV spectrum can lead to more unwanted by-products (Hofman-Caris et al., 2015).

#### 32 Concentration of reagent (H<sub>2</sub>O<sub>2</sub>)(Ozone, UV)

The addition of H<sub>2</sub>O<sub>2</sub> to ozone or UV treatment increases the production of OH radicals, and thus the removal efficiency of the treatment (Ocampo-Pérez et al., 2010; Real et al., 2009; Ternes et al., 2003; Zwiener and Frimmel, 2000). An increase of the H<sub>2</sub>O<sub>2</sub> dose can be used to compensate for NOM scavenging (Zwiener and Frimmel, 2000).

#### 33 Fouling (NF/RO membranes)

Fouling and the type of fouling is important for the evaluation of the removal efficiency of the NF/RO membrane. Organic fouling/biofouling has been referred to under 14 *NOM concentration (and composition)*. Fouling with NOM can lead to higher removal of negatively charged CECs and lower removal of positively charged CECs. Pressure and flux is also influenced by fouling (added resistance) as it takes more pressure to force water through a fouled membrane while keeping the flux stable, see also 34–35 *Transmembrane pressure and permeate flux* (Taheran et al., 2016).

34-35 *Transmembrane pressure and permeate flux (NF/RO membranes)*

The transmembrane pressure (TMP) is related to the permeate flux in a NF/RO membrane, the higher the transmembrane pressure the higher the flux, when the resistance is constant (Plakas and Karabelas, 2012). Increased pressure lead to higher removal of

CECs (Plakas and Karabelas, 2012). The highest removal efficiency of pesticides was found with NF membranes operation with a high flux by Chen et al. (2004) and Ahmad et al. (2008). In the 49 evaluated papers TMP was not always given. Instead the working pressure, the applied pressure or just pressure was given, obstructing comparison across studies.

#### 36 Cross-flow velocity (NF membranes)

During membrane filtration the phenomenon of concentration polarization occurs, resulting in a higher concentration of solutes in front of the membrane surface, resulting in a lower rejection of solutes compared to the bulk solution (Hajibabania et al., 2011a; Ng and Elimelech, 2004). A higher cross-flow velocity results in a lower concentration polarization, and therefore a higher CEC rejection (Crittenden and Montgomery Watson, 2012). The cross flow velocity cleans the membrane, the higher the velocity the more fouling (resistance) is removed (Crittenden and Montgomery Watson, 2012). The influence of fouling on removal of CECs has been discussed under 33 Fouling.

#### 37 Backwashing (GAC)

Backwashing can influence the CEC removal of the GAC. It removes materials blocking the pores of the GAC enhancing the removal efficiency on the other hand it can also disturb the build-up of the bed and remove organic material degrading the CECs which will decrease the removal efficiency, see also 29 Biological activity and 14 NOM concentration (and composition). It is therefore important to state if backwashing is done and how it has been done.

#### 38 Recovery (NF/RO membranes)

The recovery of a NF/RO membrane has an effect on the removal efficiency (Bellona et al., 2004). Several studies have found that removal efficiency of CECs decreases with increasing recovery (Chellam and Taylor, 2001; Chen et al., 2004).

#### 39 MWCO and/or salt rejection data (NF/RO membranes)

The rejection capacity of NF/RO membranes can be characterized by the salt passage or rejection of standard salts. For RO membranes typically monovalent salts, such as NaCl are used, while for NF membranes bivalent salts such as MgSO<sub>4</sub> are used (Bellona et al., 2004; Kiso et al. 1992, 1996, 2001). It is preferable that these characteristics are noted in the original study as there might be minor changes from batch to batch, producers and production processes change. The molecular weight cut-off (MWCO) is also used in NF/RO, referring to the lowest molecular weight solute (in daltons) in which 90% of the solute is retained by the membrane. The MWCO, however, can be difficult to obtain, and is due to differences in protocols used by manufacturers not always comparable between NF/RO membranes (Bellona et al., 2004; Cleveland et al., 2002).

### Experimental conditions

All experimental conditions are essential to report as they are all unique to the specific treatment study, and can influence the removal efficiency of the studied technique.

#### 40 Scale of experiment

As differences can be expected between lab- pilot- and full scale experiments, it is important to know the scale of the study (Hofman-Caris et al., 2017; Verliefde et al., 2009b). Lab scale refers to laboratory scale experiments (proof of principle). This can be either batch experiments (batch volume < 100 L) or continuous flow experiments (flow < 100 L/h). Pilot scale (proof of practice) refers to experiments on a larger scale, often on-site, with a flow rate normally < 10 m<sup>3</sup>/h. These scales proceed full scale application (proof of market).

#### 41 Stand alone or part of a treatment train

When the study is done on a treatment train it is important to know the impact parameters of the water matrix, and the removal efficiency of the CECs at each step of the treatment train, at least if the authors want to conclude anything on the individual treatment

steps. In all studies it is relevant to know whether the tested water has received any form of pre-treatment in advance of the treatment technology investigated in order to have an understanding of what to expect in the tested water matrix, see also 12 Type of water. The impact parameters of the water matrix should be measured after the pre-treatment. In case any post treatment is needed it can be relevant to mention this, depending on the purpose of the study.

#### 42 CEC already present or spiked

Depending on the purpose of the study spiking can be necessary to ensure that an accurate removal percentage can be calculated. When spiking a factor 100 above the limit of quantification, a removal percentage of 99% can be determined. However, very high spiking at environmentally unrealistic concentrations may have an effect on removal efficiencies and may alter the behaviour of the CEC, which is undesirable.

#### 43 Single compound or mixture

This gives an indication of the boundaries of the study, as in reality the compounds will almost always be present in a mixture.

#### 44-45 Initial and end concentration of CECs

The initial concentration and the end concentration is particularly important in case no explanation is given on how removal efficiency percentage have been calculated.

#### 46 Sample collection

It is relevant to report how the samples have been collected, e.g. grab sampling or continuously sampling. Also the number of samples and over what time period is important to mention, in order to get an understanding of the representativeness of the samples.

#### 47 Analytical method CECs

Here it is essential to describe the analytical methods used and the treatment of the samples. Furthermore, it is important to give the limit of detection for each CEC specifically, and if possible the recoveries too. This information is very relevant as it gives a detailed understanding of what the removal efficiencies are based on. This can all be added in supplementary information.

### Reproducibility and uncertainties

To ensure the clarity and plausibility of the experimental findings, the following information (criteria 48-51) should be available to assess the reproducibility and uncertainties of the study.

#### 48 Calculation of removal efficiency

It should be clear how the removal efficiencies are calculated, and how values < LOQ/LOD are treated, as these choices highly impact the results (Helsel, 2005; Weltje and Sumpter, 2017). When spiking the CECs this problem can be avoided, by spiking a high enough concentration. When not spiking, When not spiking a '>' percentage using the LOQ/LOD as end concentration is preferred to stating a 100% removal, as this in cases with a low initial concentration and a high LOQ/LOD can be very misleading. It is also important to note the number of replicates, and how they are used in the determination of the removal efficiencies.

#### 49 Statistical methods

Have appropriate statistical techniques been employed to evaluate variability and uncertainty? If any statistics are used to analyse the experimental findings this should be mentioned including which methods and why.

#### 50 Significance

When significance of the experimental findings is presented it should be noted how this is determined and at what p-value a result is deemed significant.

#### 51 Uncertainties data

A description of the produced amount of data and to what degree it is sufficient to support the conclusions of the study is very helpful to evaluate the reliability of the results. To what extent do the uncertainty and variability of the different measurements impact the conclusions that can be inferred from the data and the

utility of the study? What is the standard deviation of the results?

## 5. Application of the criteria

The current status of reporting is discussed based on the review of 244 selected papers (Supplementary Information 1), and the benefits of a standardised framework are elaborated based on the findings. Because of the broad selection of techniques and resulting large amount of data, the evaluation of the CEC removal itself will be presented in a follow-up study.

### 5.1. Relevance criteria

Table 4 details the results of the application of the relevance criteria described in chapter 4.2.1 on 244 peer-reviewed scientific articles on the removal of CECs with one of the selected water treatment techniques.

It is important to realise that the failure to fulfil one of the relevance criteria does not necessarily mean that the paper is of poor quality, but merely that the paper does not comply with the specified relevance criteria and cannot be used for our purpose, i.e. decision-making by stakeholders comparing removal efficiencies between various water treatment techniques. It is however remarkable that so few of the evaluated papers are suitable for a comparison across techniques, compounds and water matrixes. In general the first three criteria are the ones most frequently not fulfilled. The papers were either outside the scope we were interested in, did not include removal percentages at all or did not report removal percentages in a way that could be used for our purpose. Overall, only 20% of the papers evaluated were in compliance with the relevance criteria and could be used for our purpose.

### 5.2. Reliability criteria

The application of the relevance criteria left 49 studies on which the reliability criteria could be applied. Table 5 details the results of the application of the 51 reliability criteria to the water treatment technology papers.

From Table 5, it is clear that some criteria (in bold) such as compound name, technology/combinations, wavelength, scale of experiment, whether it is a stand-alone, whether the compound is spiked or not, and the pressure for NF/RO membranes are always reported. For compound name, technology/combinations and scale of the experiment this is logical as this is also included in the relevance criteria, and the studies have been selected based on this information.

Several criteria are almost always reported (criteria 9, 12, 13, 14, 21, 27, 30, 32, 36, 39, 43, 44 and 47) however for some of these parameters such as NOM content and dosage of O<sub>3</sub> and UV, what is reported is not consistent. NOM was found to be reported in DOC, TOC and/or COD, and O<sub>3</sub> as mg/L or mg O<sub>3</sub>/g DOC which cannot easily be compared if DOC is not given.

The criteria for the target compound properties (criteria 4–10) are poorly reported, varying from not at all to 50% of the studies, except for molecular weight which is reported in 21 of 24 studies. In general these criteria are not essential to report as they can be found elsewhere. However, they facilitate the interpretation of the results in the studies themselves.

The supplier of the CECs tested (criterion 11) is in general reported, however the purity of the compound (also criterion 11) only in approximately 60% of the studies.

The criteria concerning the water matrix (criteria 12–19) except water type, pH and NOM concentration (and composition) are also not well reported, with the most reported in 50% of the studies and the least in 25%. This is essential information specific for the study, and cannot be found elsewhere.

For the treatment process conditions (criteria 20–39) not yet mentioned, reporting is also not complete. In general, they are reported in less than 40% of the studies, this information is also specific for the study and cannot be found elsewhere.

With regards to the experimental conditions (criteria 42–48), sample collection and concentration after treatment are rarely reported. Especially the lack of end concentration makes it very difficult to understand and reproduce the calculation of removal efficiencies.

Understanding removal efficiencies is even more difficult as LOD/LOQ and recoveries for the analytical methods are not always reported (criterion 47), especially when the method used to calculate the efficiencies, the number of replicas, the standard deviation and how values below the LOD/LOQ are dealt with are reported in as little as 15%–47% of the studies. This makes it almost impossible to validate the findings of the studies. There is also virtually no reporting on whether statistics have been used in the studies (criterion 49) and how. This together with details on sample collection (criterion 46) are the most underreported areas found in the studies.

No papers fulfilled all the criteria. The GAC papers fulfil 36–50% of the criteria, PAC 41–73%, ozone ( $\pm$ H<sub>2</sub>O<sub>2</sub>) 38–74%, UV ( $\pm$ H<sub>2</sub>O<sub>2</sub>) 33–64%, NF/RO 39–71%. This does not mean that the papers cannot be used, but that not all the information needed for our evaluation is available. This will make the assessment of the efficiency of the technique less straightforward as impact parameters that can

**Table 4**

Details of relevance criteria evaluated for 244 peer-reviewed scientific articles. Since the papers often investigate more than one type of treatment, the total is not the sum of all the treatments evaluated but of all the papers.

Criteria applied	Description	GAC	PAC	O <sub>3</sub> $\pm$ H <sub>2</sub> O <sub>2</sub>	UV $\pm$ H <sub>2</sub> O <sub>2</sub>	NF	RO	Total
Total number of papers evaluated		32	40	64	82	62	29	244
1	Outside scope	1	10	8	14	12	4	40
2	Without removal percentage	25	17	22	20	5	2	75
2	Data unusable*	3	6	18	29	30	16	79
3	Not appropriate compound tested	0	0	0	0	0	0	0
4	No properties of compounds	0	0	0	0	0	0	0
5	No concentration of compound	0	0	0	0	0	0	0
6	No type of water matrix	0	0	0	0	1	0	1
7	No properties of water matrix	0	0	0	0	0	0	0
8	No properties of treatment technology	0	0	1	1	1	1	2
9	No scale of experiment	0	0	0	0	0	0	0
<b>Papers fulfilling all the criteria</b>		<b>3</b>	<b>7</b>	<b>15</b>	<b>18</b>	<b>13</b>	<b>6</b>	<b>49</b>

\* Data presented as > or < values out of the chosen range, only graphs no tables.

**Table 5**  
 Details of evaluated reliability criteria. The numbers in the cells are the number of studies providing that information. If information on the criterion can be given in more than one way, or the criterion includes more than one parameter the number of studies reporting this parameter are separated with/and reported in the same order as the parameters of the criterion. When the cell is empty this criterion is not relevant for that specific water treatment technology. Numbers in ( ) are the amount of studies for which the criteria are applicable, this is only given if the criteria are not applicable to all 49 studies. Criteria in *italic* are not essential as this information can be found elsewhere, when compound name is known. Criteria in **bold** are criteria that are reported in all studies.

Criterion	GAC	PAC	O <sub>3</sub> ± H <sub>2</sub> O <sub>2</sub>	UV ± H <sub>2</sub> O <sub>2</sub>	NF	RO	Total
# of papers	3	7	15	18	13	6	49
1 Guidelines	1	1	3	3	8	4	15
2 GLP conditions	0	0	0	0	0	0	0
3 Validity	1	2	5	1	6	4	13
<b>4 Compound name</b>	<b>3</b>	<b>7</b>	<b>15</b>	<b>18</b>	<b>13</b>	<b>6</b>	<b>49</b>
5 CAS number	0	2	5	4	3	2	12
6 Form tested	0	0	0	0	0	0	0
7 Molecular charge or pKa	0	2	2	1	7	3	12
8 Log K <sub>OW</sub> (Log D <sub>ow</sub> , Log P)	1	4	4	4	12 (incl. 3 Log P)	6	22
9 Molecular weight/size	1/0	2/1			13/10	6/5	21/12(24)
10 Functional groups/reactivity	0/0	3/1	1/1	1/1	0/0	0/0	3/1
11 Supplier/purity*	2/0(2)	1/1(3)	4/3(8)	10/9(13)	6/5(13)	2/2(6)	22/15(35)
12 Water type	3	7	15	18	11	6	47
13 pH	3	4	15	17	9	5	41
14 NOM concentration (DOC,TOC,COD)**/composition	2/0(3)	4/2(6)	12(14)/(N/A)	11(13)/(N/A)	6/2(8)	2/1(4)	31/4(37)
15 Temperature	2	4	8	10	9	3	28
16 Nitrite or nitrate**			6(15)	6(13)			11(25)
17 Bromide or bromate**			4(15)				4(15)
18 Turbidity**				1(13)	3(8)	1(4)	4(21)
19 UV transmittance**				3(13)			3(13)
<b>20 Technology and combinations</b>	<b>3</b>	<b>7</b>	<b>15</b>	<b>18</b>	<b>13</b>	<b>6</b>	<b>49</b>
21 Surface or membrane area	3	6			12	5	21(24)
22 Pore volume	1	1					2(10)
23 Column length	1						1(3)
24 Flow through	1						1(3)
25 Prior use	1						1(3)
26 Surface charge	0	1			5	2	6(24)
27 Concentration/dosage/intensity		6	14	16			31(34)
28 Contact, exposure or irradiation time	3	6	13	13			27(35)
29 Biological activity	1						1(3)
30 Technology application	3	6	13	15	12	6	44
<b>31 Wavelength</b>				<b>18</b>			<b>18(18)</b>
32 Concentration of reagent (H <sub>2</sub> O <sub>2</sub> )***			6(6)	13(15)			18(20)
33 Fouling					8	3	9(14)
<b>34 Pressure</b>					<b>13</b>	<b>6</b>	<b>14(14)</b>
35 Cross-flow velocity					6	3	6(14)
36 Permeate flux					12	6	13(14)
37 Backwashing	1						1(3)
38 Recovery					6	3	7(14)
39 MWCO/salt rejection					12	5	13(14)
<b>40 Scale of experiment</b>	<b>3</b>	<b>7</b>	<b>15</b>	<b>18</b>	<b>13</b>	<b>6</b>	<b>49</b>
<b>41 Stand-alone/treatment train</b>	<b>3</b>	<b>7</b>	<b>15</b>	<b>18</b>	<b>13</b>	<b>6</b>	<b>49</b>
<b>42 Compounds spiked or not</b>	<b>3</b>	<b>7</b>	<b>15</b>	<b>18</b>	<b>13</b>	<b>6</b>	<b>49</b>
43 Single compound or mixture	3	7	15	17	13	6	48
44 Initial concentrations	3	7	15	14	10	6	43
45 End concentrations	0	2	7	2	0	0	9
46 Collection of sample	0	1	4	0	1	0	6
47 Analytical method/recoveries/(LOD/LOQ)	3/1/3	7/3/7	14/6/12	18/2/7	13/3/9	6/1/3	48/11/31
48 Calculation of removal efficiency/replicas/use of LOD/LOQ	0/1/0	2/4/1	7/11/5	4/4/1	11/6/0	5/3/0	23/20/7
49 Statistical methods	0	2	3	1	2	1	7
50 Statistically significant	0	2	1	1	2	1	5
51 Uncertainty data/SD	0/0	1/2	1/5	1/2	2/5	0/2	5/12

\* In 14 studies unspiked wastewater was used so supplier and purity of the compounds is in this case not applicable. \*\*12 studies are only using ultra-pure, Milli Q or distilled water, for these waters the matrix properties are known and not essential to report. \*\*\*6 O<sub>3</sub> studies and 15 UV studies applied H<sub>2</sub>O<sub>2</sub>, in this case concentration of reagent is applicable.

influence the removal efficiency are not reported and/or removal efficiencies have not been described.

## 6. Conclusions and recommendations

Our literature survey revealed that there is currently no uniform

approach to study the CEC removal efficiency of advanced water treatment technologies. Removal is tested at various scales, with different water matrices and numerous CECs. There are large variations in the process conditions of the treatment, in the experimental set-ups and in the way results are expressed and reported.

These variations hinder the interpretation and evaluation of the

removal efficiency data. This in turn makes it difficult for stakeholders to make an informed decision with regards to which treatment technology will be relevant for their specific needs. Therefore, in this study a framework for evaluation of scientific and technical information that describes what is important when evaluating removal efficiency studies was developed.

For this framework, two criteria sets were developed: 9 relevance criteria and 51 reliability criteria. These two criteria sets offer a thorough, unbiased and standardised method to select studies to evaluate and compare the CEC removal efficiency of advanced water treatment technologies in a scientifically sound way.

The relevance criteria have been applied to 244 treatment technology studies, and 49 of these papers fulfilled the criteria, with non-compliance with the criteria *outside scope* and *appropriate data* being the main reason for the papers being discarded. Especially the lack of removal percentage, and data reported in a way that it cannot be used, affect the potential of these studies to provide additional information and be suitable for a comparison across techniques, compounds and water matrixes.

The reliability criteria were applied to the 49 remaining papers. The main finding here was a severe lack of information on how removal percentages had been calculated in terms of how LOD/LOQs had been dealt with, how many replicas had been used, what statistics had been applied, and what the standard deviation was. This hinders the interpretation of the reported result. Furthermore, many papers did not report on all the identified impact parameters which makes comparison across studies difficult.

These findings clearly demonstrates the need for a more uniform approach. When the developed framework is used as a guideline by scientists to document their studies it will facilitate the comparison of different treatment technologies regarding the removal of CECs. This in turn will enhance the interpretation of the findings of the studies and consequently benefit the selection of appropriate technologies by water managers and other stakeholders.

## Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.watres.2019.05.088>.

## References

- Ågerstrand, M., Breitholtz, M., Rüdén, C., 2011a. Comparison of four different methods for reliability evaluation of ecotoxicity data: a case study of non-standard test data used in environmental risk assessments of pharmaceutical substances. *Environ. Sci. Eur.* 23 (1).
- Ågerstrand, M., Küster, A., Bachmann, J., Breitholtz, M., Ebert, I., Rechenberg, B., Rüdén, C., 2011b. Reporting and evaluation criteria as means towards a transparent use of ecotoxicity data for environmental risk assessment of pharmaceuticals. *Environ. Pollut.* 159 (10), 2487–2492.
- Ahmad, A.L., Tan, L.S., Shukor, S.R.A., 2008. Dimethoate and atrazine retention from aqueous solution by nanofiltration membranes. *J. Hazard Mater.* 151 (1), 71–77.
- ASTM, 2014a. Standard Practice for the Prediction of Contaminant Adsorption on GAC in Aqueous Systems Using Rapid Small-Scale Column Tests. ASTM International, West Conshohocken, PA, p. 6.
- ASTM, 2014b. Standard Practice for Determination of Adsorptive Capacity of Activated Carbon by Aqueous Phase Isotherm Technique. ASTM International, West Conshohocken, PA, p. 4.
- ASTM, 2014c. Standard Test Methods for Operating Characteristics of Reverse Osmosis and Nanofiltration Devices. ASTM International, West Conshohocken, PA.
- Avisar, D., Lester, Y., Mamane, H., 2010. pH induced polychromatic UV treatment for the removal of a mixture of SMX, OTC and CIP from water. *J. Hazard Mater.* 175 (1–3), 1068–1074.
- Bellona, C., Drewes, J.E., Xu, P., Amy, G., 2004. Factors affecting the rejection of organic solutes during NF/RO treatment - a literature review. *Water Res.* 38 (12), 2795–2809.
- Benami, M., Gillor, O., Gross, A., 2016. Potential health and environmental risks associated with onsite greywater reuse: a review. *Built. Environ.* 42 (2), 212–229.
- Bernhardt, E.S., Rosi, E.J., Gessner, M.O., 2017. Synthetic chemicals as agents of global change. *Front. Ecol. Environ.* 15 (2), 84–90.
- Boehler, M., Zwickenpflug, B., Hollender, J., Ternes, T., Joss, A., Siegrist, H., 2012. Removal of micropollutants in municipal wastewater treatment plants by powder-activated carbon. *Water Sci. Technol.* 66 (10), 2115–2121.
- Borowska, E., Bourgin, M., Hollender, J., Kienle, C., McArdell, C.S., von Gunten, U., 2016. Oxidation of cetirizine, fexofenadine and hydrochlorothiazide during ozonation: kinetics and formation of transformation products. *Water Res.* 94, 350–362.
- Bourgin, M., Beck, B., Boehler, M., Borowska, E., Fleiner, J., Salhi, E., Teichler, R., von Gunten, U., Siegrist, H., McArdell, C.S., 2018. Evaluation of a full-scale wastewater treatment plant upgraded with ozonation and biological post-treatments: abatement of micropollutants, formation of transformation products and oxidation by-products. *Water Res.* 129, 486–498.
- Braeken, L., Ramaekers, R., Zhang, Y., Maes, G., Van Der Bruggen, B., Vandecasteele, C., 2005. Influence of hydrophobicity on retention in nanofiltration of aqueous solutions containing organic compounds. *J. Membr. Sci.* 252 (1–2), 195–203.
- Braghetta, A., DiGianno, F.A., Ball, W.P., 1997. Nanofiltration of natural organic matter: pH and ionic strength effects. *J. Environ. Eng.* 123 (7), 628–641.
- Bruce, G.M., Pleus, R.C., Snyder, S.A., 2010. Toxicological relevance of pharmaceuticals in drinking water. *Environ. Sci. Technol.* 44 (14), 5619–5626.
- CEFIC, 2016. Facts and Figures 2016 of the European Chemical Industry CEFIC.
- Chellam, S., Taylor, J.S., 2001. Simplified analysis of contaminant rejection during ground- and surface water nanofiltration under the information collection rule. *Water Res.* 35 (10), 2460–2474.
- Chen, P.J., Kullman, S.W., Hinton, D.E., Linden, K.G., 2007. Comparisons of polychromatic and monochromatic UV-based treatments of bisphenol-A in water via toxicity assessments. *Chemosphere* 68 (6), 1041–1049.
- Chen, S.S., Taylor, J.S., Mulford, L.A., Norris, C.D., 2004. Influences of molecular weight, molecular size, flux, and recovery for aromatic pesticide removal by nanofiltration membranes. *Desalination* 160 (2), 103–111.
- Chowdhury, Z.K., 2013. Activated Carbon: Solutions for Improving Water Quality. American Water Works Association.
- Cleveland, C.T., Seacord, T.F., Zander, A.K., 2002. Standardized membrane pore size characterization by polyethylene glycol rejection. *J. Environ. Eng.* 128 (5), 399–407.
- Comerton, A.M., Andrews, R.C., Bagley, D.M., 2009. The influence of natural organic matter and cations on the rejection of endocrine disrupting and pharmaceutically active compounds by nanofiltration. *Water Res.* 43 (3), 613–622.
- Comerton, A.M., Andrews, R.C., Bagley, D.M., Hao, C., 2008. The rejection of endocrine disrupting and pharmaceutically active compounds by NF and RO membranes as a function of compound and water matrix properties. *J. Membr. Sci.* 313 (1–2), 323–335.
- Crittenden, J.C., Montgomery Watson, H., 2012. *MWH's Water Treatment: Principles and Design*. John Wiley & Sons, Hoboken, N.J.
- De Ridder, D.J., Verliefdé, A.R.D., Heijman, S.G.J., Verberk, J.Q.J.C., Rietveld, L.C., Van Der Aa, L.T.J., Amy, G.L., Van Dijk, J.C., 2011. Influence of natural organic matter on equilibrium adsorption of neutral and charged pharmaceuticals onto activated carbon. *Water Sci. Technol.* 63 (3), 416–423.
- de Ridder, D.J., Villacorte, L., Verliefdé, A.R., Verberk, J.Q., Heijman, S.G., Amy, G.L., van Dijk, J.C., 2010. Modeling equilibrium adsorption of organic micropollutants onto activated carbon. *Water Res.* 44 (10), 3077–3086.
- Deshmukh, S.S., Childress, A.E., 2001. Zeta potential of commercial RO membranes: influence of source water type and chemistry. *Desalination* 140 (1), 87–95.
- Dodd, M.C., Zuleeg, S., Von Gunten, U., Pronk, W., 2008. Ozonation of source-separated urine for resource recovery and waste minimization: process modeling, reaction chemistry, and operational considerations. *Environ. Sci. Technol.* 42 (24), 9329–9337.
- ECETOC, 2009. Framework for the Integration of Human and Animal Data in Chemical Risk Assessment. European Centre for Ecotoxicology and Toxicology of Chemicals, Brussels, Belgium.
- ECHA, 2011. Guidance on Information Requirements and Chemical Safety

- Assessment. Chapter R.4: Evaluation of available information, ECHA, Helsinki, Finland.
- EPA, U., 2012. Guidance for Evaluating and Documenting the Quality of Existing Scientific and Technical Information, Peer Review Advisory Group. Science and Technology Policy Council, US Environmental Protection Agency, Washington, DC.
- Fischer, A., ter Laak, T., Bronders, J., Desmet, N., Christoffels, E., van Wezel, A., van der Hoek, J.P., 2017. Decision support for water quality management of contaminants of emerging concern. *J. Environ. Manag.* 193, 360–372.
- Hagmeyer, G., Gimbel, R., 1998. Modelling the salt rejection of nanofiltration membranes for ternary ion mixtures and for single salts at different pH values. *Desalination* 117 (1–3), 247–256.
- Hajibabania, S., Verliefde, A., Drewes, J.E., Nghiem, L.D., McDonald, J., Khan, S., Le-Clech, P., 2011a. Effect of fouling on removal of trace organic compounds by nanofiltration. *Drink. Water Eng. Sci.* 4 (1), 71–82.
- Hajibabania, S., Verliefde, A., McDonald, J.A., Khan, S.J., Le-Clech, P., 2011b. Fate of trace organic compounds during treatment by nanofiltration. *J. Membr. Sci.* 373 (1–2), 130–139.
- Helsel, D.R., 2005. *Nondetects and Data Analysis: Statistics for Censored Environmental Data*. John Wiley & Sons.
- Hofman-Caris, C.H.M., Siegers, W.G., van de Merlen, K., de Man, A.W.A., Hofman, J.A.M.H., 2017. Removal of pharmaceuticals from WWTP effluent: removal of EfOM followed by advanced oxidation. *Chem. Eng. J.* 327, 514–521.
- Hofman-Caris, R.C.H.M., Harmsen, D.J.H., Puijker, L., Baken, K.A., Wols, B.A., Beerendonk, E.F., Keltjens, L.L.M., 2015. Influence of process conditions and water quality on the formation of mutagenic byproducts in UV/H<sub>2</sub>O<sub>2</sub> processes. *Water Res.* 74, 191–202.
- Hollender, J., Zimmermann, S.G., Koepke, S., Krauss, M., McArdell, C.S., Ort, C., Singer, H., Von Gunten, U., Siegrist, H., 2009. Elimination of organic micropollutants in a municipal wastewater treatment plant upgraded with a full-scale post-ozonation followed by sand filtration. *Environ. Sci. Technol.* 43 (20), 7862–7869.
- Hu, J.Y., Aizawa, T., Ookubo, Y., Morita, T., Magara, Y., 1998. Adsorptive characteristics of ionogenic aromatic pesticides in water on powdered activated carbon. *Water Res.* 32 (9), 2593–2600.
- Hu, J.Y., Jin, X., Ong, S.L., 2007. Rejection of estrone by nanofiltration: influence of solution chemistry. *J. Membr. Sci.* 302 (1–2), 188–196.
- Isigonis, P., Ciffroy, P., Zabeo, A., Semenzin, E., Critto, A., Giove, S., Marcomini, A., 2015. A Multi-Criteria Decision Analysis based methodology for quantitatively scoring the reliability and relevance of ecotoxicological data. *Sci. Total Environ.* 538, 102–116.
- ISO, 2017. *Standards Catalogue*.
- Jeirani, Z., Niu, C.H., Soltan, J., 2017. Adsorption of emerging pollutants on activated carbon. *Rev. Chem. Eng.* 33 (5), 491–522.
- Jung, Y.J., Kiso, Y., Adawih binti Othman, R.A., Ikeda, A., Nishimura, K., Min, K.S., Kumano, A., Arijai, A., 2005. Rejection properties of aromatic pesticides with a hollow-fiber NF membrane. *Desalination* 180 (1–3), 63–71.
- Kase, R., Korkaric, M., Werner, I., Ågerstrand, M., 2016. Criteria for Reporting and Evaluating ecotoxicity Data (CRED): comparison and perception of the Klimisch and CRED methods for evaluating reliability and relevance of ecotoxicity studies. *Environ. Sci. Eur.* 28 (1), 1–14.
- Kilduff, J.E., Karanfil, T., Weber Jr., W.J., 1998. Competitive effects of nondisplaceable organic compounds on trichloroethylene uptake by activated carbon. II. Model verification and applicability to natural organic matter. *J. Colloid Interface Sci.* 205 (2), 280–289.
- Kim, S.L., Paul Chen, J., Ting, Y.P., 2002. Study on feed pretreatment for membrane filtration of secondary effluent. *Separ. Purif. Technol.* 29 (2), 171–179.
- Kiso, Y., Kitao, T., Jinno, K., Miyagi, M., 1992. The effects of molecular width on permeation of organic solute through cellulose acetate reverse osmosis membranes. *J. Membr. Sci.* 74 (1–2), 95–103.
- Kiso, Y., Li, H., Kitao, T., 1996. Pesticides separation by nanofiltration membranes. *J. Jpn. Soc. Water Environ.* 10, 648–656.
- Kiso, Y., Sugiura, Y., Kitao, T., Nishimura, K., 2001. Effects of hydrophobicity and molecular size on rejection of aromatic pesticides with nanofiltration membranes. *J. Membr. Sci.* 192 (1–2), 1–10.
- Klimisch, H.J., Andreae, M., Tillmann, U., 1997. A systematic approach for evaluating the quality of experimental toxicological and ecotoxicological data. *Regul. Toxicol. Pharmacol.* 25 (1), 1–5.
- Košutić, K., Kunst, B., 2002. Removal of organics from aqueous solutions by commercial RO and NF membranes of characterized porosities. *Desalination* 142 (1), 47–56.
- Kovalova, L., Siegrist, H., Von Gunten, U., Eugster, J., Hagenbuch, M., Wittmer, A., Moser, R., McArdell, C.S., 2013. Elimination of micropollutants during post-treatment of hospital wastewater with powdered activated carbon, ozone, and UV. *Environ. Sci. Technol.* 47 (14), 7899–7908.
- Kruithof, J.C., Martijn, B.J., 2013. UV/H<sub>2</sub>O<sub>2</sub> treatment: an essential process in a multi barrier approach against trace chemical contaminants. *Water Sci. Technol. Water Supply* 13 (1), 130–138.
- Lee, Y., Gerrity, D., Lee, M., Bogeat, A.E., Salhi, E., Gamage, S., Trenholm, R.A., Wert, E.C., Snyder, S.A., von Gunten, U., 2013. Prediction of micropollutant elimination during ozonation of municipal wastewater effluents: use of kinetic and water specific information. *Environ. Sci. Technol.* 47 (11), 5872–5881.
- Lee, Y., von Gunten, U., 2010. Oxidative transformation of micropollutants during municipal wastewater treatment: comparison of kinetic aspects of selective (chlorine, chlorine dioxide, ferrateVI, and ozone) and non-selective oxidants (hydroxyl radical). *Water Res.* 44 (2), 555–566.
- Leech, D.M., Snyder, M.T., Wetzel, R.G., 2009. Natural organic matter and sunlight accelerate the degradation of 17β-estradiol in water. *Sci. Total Environ.* 407 (6), 2087–2092.
- Lester, Y., Gozlan, I., Avisar, D., Mamane, H., 2008. Photodegradation of Sulphadimethoxine in Water by Medium Pressure UV Lamp, pp. 1147–1154.
- Liu, B., Liu, X., 2004. Direct photolysis of estrogens in aqueous solutions. *Sci. Total Environ.* 320 (2–3), 269–274.
- Loos, R., Carvalho, R., António, D.C., Comero, S., Locoro, G., Tavazzi, S., Paracchini, B., Ghiani, M., Lettieri, T., Blaha, L., Jarosova, B., Voorspoels, S., Servaes, K., Haglund, P., Fick, J., Lindberg, R.H., Schwesig, D., Gawlik, B.M., 2013. EU-wide monitoring survey on emerging polar organic contaminants in wastewater treatment plant effluents. *Water Res.* 47 (17), 6475–6487.
- Loos, R., Gawlik, B.M., Locoro, G., Rimaviciute, E., Contini, S., Bidoglio, G., 2009. EU-wide survey of polar organic persistent pollutants in European river waters. *Environ. Pollut.* 157 (2), 561–568.
- Loos, R., Locoro, G., Comero, S., Contini, S., Schwesig, D., Werres, F., Balsaa, P., Gans, O., Weiss, S., Blaha, L., Bolchi, M., Gawlik, B.M., 2010. Pan-European survey on the occurrence of selected polar organic persistent pollutants in ground water. *Water Res.* 44 (14), 4115–4126.
- Luo, Y., Guo, W., Ngo, H.H., Nghiem, L.D., Hai, F.I., Zhang, J., Liang, S., Wang, X.C., 2014. A review on the occurrence of micropollutants in the aquatic environment and their fate and removal during wastewater treatment. *Sci. Total Environ.* 473–474, 619–641.
- Magic-Knezev, A., van der Kooij, D., 2004. Optimisation and significance of ATP analysis for measuring active biomass in granular activated carbon filters used in water treatment. *Water Res.* 38 (18), 3971–3979.
- Mailler, R., Gasperi, J., Coquet, Y., Derome, C., Bulet, A., Vulliet, E., Bressy, A., Varrault, G., Chebbo, G., Rocher, V., 2016. Removal of emerging micropollutants from wastewater by activated carbon adsorption: experimental study of different activated carbons and factors influencing the adsorption of micropollutants in wastewater. *J. Chem. Environ. Eng.* 4 (1), 1102–1109.
- Mailler, R., Gasperi, J., Coquet, Y., Deshayes, S., Zedek, S., Cren-Olivé, C., Cartiser, N., Eudes, V., Bressy, A., Caupos, E., Moilleron, R., Chebbo, G., Rocher, V., 2014. Study of a large scale powdered activated carbon pilot: removals of a wide range of emerging and priority micropollutants from wastewater treatment plant effluents. *Water Res.* 72, 315–330.
- Mamy, L., Patureau, D., Barriuso, E., Bedos, C., Bessac, F., Louchart, X., Martin-Laurent, F., Miegé, C., Benoit, P., 2015. Prediction of the fate of organic compounds in the environment from their molecular properties: a review. *Crit. Rev. Environ. Sci. Technol.* 45 (12), 1277–1377.
- Margot, J., Kienle, C., Magnet, A., Weil, M., Rossi, L., de Alencastro, L.F., Abegglen, C., Thonney, D., Chèvre, N., Schärer, M., Barry, D.A., 2013. Treatment of micropollutants in municipal wastewater: ozone or powdered activated carbon? *Sci. Total Environ.* 461–462, 480–498.
- Matsui, Y., Knappe, D.R.U., Takagi, R., 2002. Pesticide adsorption by granular activated carbon adsorbents. I. Effect of natural organic matter preloading on removal rates and model simplification. *Environ. Sci. Technol.* 36 (15), 3426–3431.
- Moermond, C.T.A., Kase, R., Korkaric, M., Ågerstrand, M., 2016. CRED: criteria for reporting and evaluating ecotoxicity data. *Environ. Toxicol. Chem.* 35 (5), 1297–1309.
- Moons, K., Van der Bruggen, B., 2006. Removal of micropollutants during drinking water production from surface water with nanofiltration. *Desalination* 199 (1–3), 245–247.
- Moreno-Castilla, C., 2004. Adsorption of organic molecules from aqueous solutions on carbon materials. *Carbon* 42 (1), 83–94.
- Nakada, N., Shinohara, H., Murata, A., Kiri, K., Managaki, S., Sato, N., Takada, H., 2007. Removal of selected pharmaceuticals and personal care products (PPCPs) and endocrine-disrupting chemicals (EDCs) during sand filtration and ozonation at a municipal sewage treatment plant. *Water Res.* 41 (19), 4373–4382.
- Nam, S.W., Choi, D.J., Kim, S.K., Her, N., Zoh, K.D., 2014. Adsorption characteristics of selected hydrophilic and hydrophobic micropollutants in water using activated carbon. *J. Hazard Mater.* 270, 144–152.
- Neamtu, M., Frimmel, F.H., 2006. Degradation of endocrine disrupting bisphenol A by 254 nm irradiation in different water matrices and effect on yeast cells. *Water Res.* 40 (20), 3745–3750.
- Newcombe, G., Drikas, M., 1997. Adsorption of NOM onto activated carbon: electrostatic and non-electrostatic effects. *Carbon* 35 (9), 1239–1250.
- Ng, H.Y., Elimelech, M., 2004. Influence of colloidal fouling on rejection of trace organic contaminants by reverse osmosis. *J. Membr. Sci.* 244 (1–2), 215–226.
- Nghiem, L.D., Coleman, P.J., 2008. NF/RO filtration of the hydrophobic ionogenic compound triclosan: transport mechanisms and the influence of membrane fouling. *Separ. Purif. Technol.* 62 (3), 709–716.
- Nghiem, L.D., Coleman, P.J., Espendiller, C., 2010. Mechanisms underlying the effects of membrane fouling on the nanofiltration of trace organic contaminants. *Desalination* 250 (2), 682–687.
- Nghiem, L.D., Hawkes, S., 2007. Effects of membrane fouling on the nanofiltration of pharmaceutically active compounds (PhACs): mechanisms and role of membrane pore size. *Separ. Purif. Technol.* 57 (1), 176–184.
- Ocampo-Pérez, R., Sánchez-Polo, M., Rivera-Utrilla, J., Leyva-Ramos, R., 2010. Degradation of antineoplastic cytarabine in aqueous phase by advanced oxidation processes based on ultraviolet radiation. *Chem. Eng. J.* 165 (2), 581–588.
- OECD, 1998. *OECD Series on Principles of Good Laboratory Practice (GLP) and*

- Compliance Monitoring. OECD, Paris, France.
- OECD, 2017. OECD Guidelines for the Testing of Chemicals.
- Ozaki, H., Li, H., 2002. Rejection of organic compounds by ultra-low pressure reverse osmosis membrane. *Water Res.* 36 (1), 123–130.
- Pelekani, C., Snoeyink, V.L., 1999. Competitive adsorption in natural water: role of activated carbon pore size. *Water Res.* 33 (5), 1209–1219.
- Pereira, V.J., Weinberg, H.S., Linden, K.G., Singer, P.C., 2007. UV degradation kinetics and modeling of pharmaceutical compounds in laboratory grade and surface water via direct and indirect photolysis at 254 nm. *Environ. Sci. Technol.* 41 (5), 1682–1688.
- Plakas, K.V., Karabelas, A.J., 2012. Removal of pesticides from water by NF and RO membranes - a review. *Desalination* 287, 255–265.
- Pradhan, S., Al-Ghamdi, S.G., Mackey, H.R., 2019. Greywater recycling in buildings using living walls and green roofs: a review of the applicability and challenges. *Sci. Total Environ.* 652, 330–344.
- Rattier, M., Reungoat, J., Keller, J., Gernjak, W., 2014. Removal of micropollutants during tertiary wastewater treatment by biofiltration: role of nitrifiers and removal mechanisms. *Water Res.* 54, 89–99.
- Real, F.J., Javier Benitez, F., Acero, J.L., Sagasti, J.J.P., Casas, F., 2009. Kinetics of the chemical oxidation of the pharmaceuticals primidone, ketoprofen, and diazotriazole in ultrapure and natural waters. *Ind. Eng. Chem. Res.* 48 (7), 3380–3388.
- Reemtsma, T., Berger, U., Arp, H.P.H., Gallard, H., Knepper, T.P., Neumann, M., Quintana, J.B., Voogt, P.D., 2016. Mind the gap: persistent and mobile organic compounds - water contaminants that slip through. *Environ. Sci. Technol.* 50 (19), 10308–10315.
- Rivera-Utrilla, J., Sánchez-Polo, M., Ferro-García, M.Á., Prados-Joya, G., Ocampo-Pérez, R., 2013. Pharmaceuticals as emerging contaminants and their removal from water. A review. *Chemosphere* 93 (7), 1268–1287.
- Rosenfeldt, E.J., Linden, K.G., 2004. Degradation of endocrine disrupting chemicals bisphenol A, ethinyl estradiol, and estradiol during UV photolysis and advanced oxidation processes. *Environ. Sci. Technol.* 38 (20), 5476–5483.
- Rossner, A., Snyder, S.A., Knappe, D.R., 2009. Removal of emerging contaminants of concern by alternative adsorbents. *Water Res.* 43 (15), 3787–3796.
- Roth, N., Ciffroy, P., 2016. A critical review of frameworks used for evaluating reliability and relevance of (eco)toxicity data: perspectives for an integrated eco-human decision-making framework. *Environ. Int.* 95, 16–29.
- Sadmani, A.H.M.A., Andrews, R.C., Bagley, D.M., 2014. Impact of natural water colloids and cations on the rejection of pharmaceutically active and endocrine disrupting compounds by nanofiltration. *J. Membr. Sci.* 450, 272–281.
- Schriks, M., Heringa, M.B., van der Kooij, M.M., de Voogt, P., van Wezel, A.P., 2010. Toxicological relevance of emerging contaminants for drinking water quality. *Water Res.* 44 (2), 461–476.
- Schwarzenbach, R.P., Escher, B.I., Fenner, K., Hofstetter, T.B., Johnson, C.A., Von Gunten, U., Wehrli, B., 2006. The challenge of micropollutants in aquatic systems. *Science* 313 (5790), 1072–1077.
- Snyder, S., Wert, E., Lei, H., Westerhoff, P., Yoon, Y., 2007a. Removal of EDCs and Pharmaceuticals in Drinking and Reuse Treatment Processes. AWWA Research Foundation.
- Snyder, S.A., Adham, S., Redding, A.M., Cannon, F.S., DeCarolis, J., Oppenheimer, J., Wert, E.C., Yoon, Y., 2007b. Role of membranes and activated carbon in the removal of endocrine disruptors and pharmaceuticals. *Desalination* 202 (1–3), 156–181.
- Taheran, M., Brar, S.K., Verma, M., Surampalli, R.Y., Zhang, T.C., Valero, J.R., 2016. Membrane processes for removal of pharmaceutically active compounds (PhACs) from water and wastewaters. *Sci. Total Environ.* 547, 60–77.
- Ternes, T.A., Stüber, J., Herrmann, N., McDowell, D., Ried, A., Kampmann, M., Teiser, B., 2003. Ozonation: a tool for removal of pharmaceuticals, contrast media and musk fragrances from wastewater? *Water Res.* 37 (8), 1976–1982.
- UNEP, 2013. Global Chemicals Outlook-Towards Sound Management of Chemicals.
- USEPA, 2003. A Summary of General Assessment Factors for Evaluating the Quality of Scientific and Technical Information. Science Policy Council, Washington DC.
- van Wezel, A.P., ter Laak, T.L., Fischer, A., Bauerlein, P.S., Munthe, J., Posthuma, L., 2017. Mitigation options for chemicals of emerging concern in surface waters: operationalising solutions-focused risk assessment. *Environ. Sci.: Water Res. Technol.* 3 (3), 403–414.
- Verlicchi, P., Al Aukidy, M., Zambello, E., 2012. Occurrence of pharmaceutical compounds in urban wastewater: removal, mass load and environmental risk after a secondary treatment-A review. *Sci. Total Environ.* 429, 123–155.
- Verlicchi, P., Galletti, A., Petrovic, M., Barceló, D., 2010. Hospital effluents as a source of emerging pollutants: an overview of micropollutants and sustainable treatment options. *J. Hydrol.* 389 (3–4), 416–428.
- Verliefde, A., Cornelissen, E., Amy, G., Van der Bruggen, B., van Dijk, H., 2007a. Priority organic micropollutants in water sources in Flanders and The Netherlands and assessment of removal possibilities with nanofiltration. *Environ. Pollut.* 146 (1), 281–289.
- Verliefde, A.R.D., Cornelissen, E.R., Heijman, S.G.J., Petrinic, I., Luxbacher, T., Amy, G.L., Van der Bruggen, B., van Dijk, J.C., 2009a. Influence of membrane fouling by (pretreated) surface water on rejection of pharmaceutically active compounds (PhACs) by nanofiltration membranes. *J. Membr. Sci.* 330 (1–2), 90–103.
- Verliefde, A.R.D., Cornelissen, E.R., Heijman, S.G.J., Verberk, J.Q.J.C., Amy, G.L., Van Der Bruggen, B., Van Dijk, J.C., 2009b. Construction and validation of a full-scale model for rejection of organic micropollutants by NF membranes. *J. Membr. Sci.* 322 (1), 52–66.
- Verliefde, A.R.D., Cornelissen, E.R., Heijman, S.G.J., Verberk, J.Q.J.C., Amy, G.L., Van der Bruggen, B., van Dijk, J.C., 2009b. Construction and validation of a full-scale model for rejection of organic micropollutants by NF membranes. *J. Membr. Sci.* 339 (1–2), 10–20.
- Verliefde, A.R.D., Heijman, S.G.J., Cornelissen, E.R., Amy, G., Van Der Bruggen, B., Van Dijk, J.C., 2007b. Charge Effects on the Rejection of Pharmaceuticals with Nanofiltration.
- von Gunten, U., 2003a. Ozonation of drinking water: Part I. Oxidation kinetics and product formation. *Water Res.* 37 (7), 1443–1467.
- von Gunten, U., 2003b. Ozonation of drinking water: Part II. Disinfection and by-product formation in presence of bromide, iodide or chlorine. *Water Res.* 37 (7), 1469–1487.
- von Sonntag, C., von Gunten, U., 2012. *Chemistry of Ozone in Water and Wastewater Treatment*. IWA Publishing.
- Weltje, L., Sumpter, J.P., 2017. What makes a concentration environmentally relevant? Critique and a proposal. *Environ. Sci. Technol.* 51 (20), 11520–11521.
- Westerhoff, P., Yoon, Y., Snyder, S., Wert, E., 2005. Fate of endocrine-disruptor, pharmaceutical, and personal care product chemicals during simulated drinking water treatment processes. *Environ. Sci. Technol.* 39 (17), 6649–6663.
- Wilsenach, J.A., Van Loosdrecht, M.C.M., 2004. Effects of separate urine collection on advanced nutrient removal processes. *Environ. Sci. Technol.* 38 (4), 1208–1215.
- Wols, B.A., Harmsen, D.J.H., van Remmen, T., Beerendonk, E.F., Hofman-Caris, C.H.M., 2015. Design aspects of UV/H2O2 reactors. *Chem. Eng. Sci.* 137, 712–721.
- Wols, B.A., Hofman-Caris, C.H.M., 2012. Review of photochemical reaction constants of organic micropollutants required for UV advanced oxidation processes in water. *Water Res.* 46 (9), 2815–2827.
- Wols, B.A., Hofman-Caris, C.H.M., Harmsen, D.J.H., Beerendonk, E.F., 2013. Degradation of 40 selected pharmaceuticals by UV/H2O2. *Water Res.* 47 (15), 5876–5888.
- Yang, W., Zhou, H., Cicek, N., 2014. Treatment of organic micropollutants in water and wastewater by UV-based processes: a literature review. *Crit. Rev. Environ. Sci. Technol.* 44 (13), 1443–1476.
- Yang, Y., Ok, Y.S., Kim, K.H., Kwon, E.E., Tsang, Y.F., 2017. Occurrences and removal of pharmaceuticals and personal care products (PPCPs) in drinking water and water/sewage treatment plants: a review. *Sci. Total Environ.* 596–597, 303–320.
- Yangali-Quintanilla, V., Sadmani, A., McConville, M., Kennedy, M., Amy, G., 2009. Rejection of pharmaceutically active compounds and endocrine disrupting compounds by clean and fouled nanofiltration membranes. *Water Res.* 43 (9), 2349–2362.
- Yoon, Y., Westerhoff, P., Snyder, S.A., Wert, E.C., 2006. Nanofiltration and ultrafiltration of endocrine disrupting compounds, pharmaceuticals and personal care products. *J. Membr. Sci.* 270 (1–2), 88–100.
- Yoon, Y., Westerhoff, P., Snyder, S.A., Wert, E.C., Yoon, J., 2007. Removal of endocrine disrupting compounds and pharmaceuticals by nanofiltration and ultrafiltration membranes. *Desalination* 202 (1–3), 16–23.
- Yuan, F., Hu, C., Hu, X., Qu, J., Yang, M., 2009. Degradation of selected pharmaceuticals in aqueous solution with UV and UV/H2O2. *Water Res.* 43 (6), 1766–1774.
- Zhan, M., Yang, X., Xian, Q., Kong, L., 2006. Photosensitized degradation of bisphenol A involving reactive oxygen species in the presence of humic substances. *Chemosphere* 63 (3), 378–386.
- Zhang, Z., Zhu, H., Wen, X., Si, X., 2012. Degradation behavior of 17 $\alpha$ -ethinylestradiol by ozonation in the synthetic secondary effluent. *J. Environ. Sci.* 24 (2), 228–233.
- Zucker, I., Avisar, D., Mamane, H., Jekel, M., Hübner, U., 2016. Determination of oxidant exposure during ozonation of secondary effluent to predict contaminant removal. *Water Res.* 100, 508–516.
- Zwiener, C., Frimmel, F.H., 2000. Oxidative treatment of pharmaceuticals in water. *Water Res.* 34 (6), 1881–1885.