Numerical analysis of TDS spectra under high and low flux plasma exposure conditions

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Abstract

Recently developed numerical model, based on the dislocation-driven nucleation of gas bubbles, is used to analyse experimental results on deuterium retention in tungsten under ITER relevant plasma exposure conditions. Focus is put on understanding the relation between exposure temperature and flux on primary features of thermal desorption spectra: peak positions and intensities of the desorption flux. The model allows one to relate the peak positions with the size of plasma induced deuterium bubbles and envisage exposure conditions (temperature and flux) for their formation. Based on the performed analysis, dedicated experimental conditions to validate the model are proposed.

Keywords: hydrogen, retention, plasma, blisters, bubbles.

1. Introduction

Development and qualification of plasma facing materials is one of the main challenges in designing large-scale fusion devices(i.e. ITER and DEMO). Due to its favorable physical properties, tungsten (W) is considered to be one of the main candidate materials and is chosen for divertor armour in ITER [1]. The latter should be able to withstand severe conditions in terms of heat and particle loads without considerable degradation of its mechanical and thermal properties within scheduled operational time. Moreover, the ITER nuclear licensing imposes limits on the amount of tritium making the problem of hydrogen isotope (hereinafter referred to as "hydrogen" or "H") retention an additional technological challenge.

H retention is expressed in surface modification, formation of subsurface blisters and accumulation of H in the bulk material. These effects are attributed to trapping of H atoms on lattice defects such as vacancies, dislocations/grain boundaries and voids [2-4]. Typical kinetic energy of H ions coming from plasma (below 500 eV) is far below the threshold energy needed to create a stable Frenkel-pair

defect in W, and the implantation range of H ions is limited to several nanometers [5]. Nevertheless, experiments with ion beams and linear plasma generators show that H penetrates up to a depth of several μ m [6-8]. H retention under such conditions cannot be assigned to vacancies since the thermal concentration of vacancies is negligible and no vacancies are created during plasma exposure directly via displacement damage. However, Thermal Desorption Spectroscopy (TDS) demonstrates three typical release stages, attributed to static traps in the material or also called "natural traps": dislocations, grain boundaries, vacancies and voids [8]. The formation of the vacancy-like trapping defects with subsequent bubble nucleation/growth and blistering under sub-threshold exposure conditions was proposed to originate from the dynamic plasma-induced material modification in the vicinity of a trapping lattice defect, namely: a screw dislocation line.

An alternative model of plasma-induced dislocation-driven bubble formation was proposed recently on the basis of *ab initio* calculations [9]. It was demonstrated that H atoms are attracted to a screw dislocation core and exhibit fast one-dimensional migration along the dislocation line. However, the binding energy of this interaction of 0.5 eV is not high enough to keep H atoms trapped at experimental temperature range (usually above 400 K). According to the calculations, once a cluster of eight H atoms is formed, it should spontaneously transform into a vacancy-like defect (jog) on a dislocation line, and this defect traps H with a considerably higher binding energy. Thus, dislocations (and dislocation junctions) are suggested to act as nucleation sites for stable H bubbles. The interaction strength of H with a newly formed jog on a dislocation is very similar to vacancy-H, as was confirmed by *ab initio* calculations in [9]. Further growth of H-jog clusters and accumulation of vacancy jogs will lead to the formation of a nanometric cavity-like defect. Theses defects will exhibit extended open volume, and thus, H interaction with such traps will be similar to nano-voids and cavities. The role of dislocations in bubble formation was validated in recent experiments involving the plasma exposure of annealed and plastically deformed W samples [9, 10]. Based on the dislocation-driven nucleation, a numerical simulation tool was developed and used to predict the conditions for the bubble formation depending on exposure temperature and flux [11]. However under the conditions where dislocationdriven mechanism is not active, other microstructural features may act as nucleation sites for bubble nucleation. For example in work [12], the observation of bimodal size distribution of blisters was attributed to an alternative mechanisms – cavity growth and coalescence on grain boundary interfaces.

In Ref [13], the effect of exposure flux - temperature on surface modification and TDS spectra was studied by a set of experiments varying the flux and exposure temperature, respectively, in the range of $9x10^{21} - 5x10^{23}$ D/m²/s and 530 - 870 K. The goal of this work is to estimate H release from H-jog and void-like (multiple vacancy jogs) defects formed during the exposure by successive jog-punching events in support of interpretation of the desorption spectra. We propose a parameterization of the binding energy, defining the H release based on *ab initio* results for H interaction with jogs and cavities as limiting cases. For describing the binding of H to jogs, and jog clusters (in transition state from jogs to bubbles), we use size-dependent fitting. By comparing the release rate, computed following our model, with experimental TDS spectra we establish a relation between positions of release stages and mean size of the defects releasing H.

2. Model description

Following the above mentioned dislocation-driven model of H trapping, one can distinguish the formation of four types of defects: (I) sub-critical H clusters (up to eight H), which are not yet capable to punch out W matrix atoms; (II) super-critical H clusters, whose transformation into jog- H_N configuration is energetically favorable; (III) large super-critical clusters, which already released several jogs for itself and grow further; and finally (IV) nanometric bubble attached to a dislocation

line, whose properties (in terms of H trapping) are equivalent to the usual bubble in W bulk. Here, we will not consider the trapping and release from the defects of type I, since these clusters are unstable above 450 K, while the latter being the lower bound of the relevant temperature range.

In work [3], a rate theory model of H desorption from tungsten surface and subsurface defects demonstrated that H release at temperatures, higher than 600 K is not affected by surface recombination. Here, we deal with assessment of high temperature desorption stage (above 600K), and therefore assume an immediate release of H from the surface after detrapping from the nano-cavities and vacancies. The main parameter of the model that defines the release rate of H from a certain type of a trapping defect is the binding energy (E_b). As mentioned before, *ab initio* calculations showed that dislocation jogs exhibit trapping properties similar to vacancies [9]. The corresponding value of the binding energy is 1.24 eV for one H atom and it decreases with an increase of H atoms trapped at the jog. The dependence of H_N -jog binding energy on N in a cluster is plotted in Fig. 1. We fitted these data points to generate a family of E_b functions for super-critical clusters. The release rate from such defects is defined as:

$$R_J^- = C_H \nu \exp(-\frac{E_b + E_m}{kT}) \qquad (1)$$

where C_H – concentration of trapped H atoms (assuming that each H in a cluster has equal probability for detrapping event), ν - Debye frequency, E_b – binding energy, E_m – migration energy (0.4 eV is taken from Frauenfelder's data [14]), k - Boltzmann constant, T – temperature.

For large vacancy clusters (i.e. nano-voids, type IV defects in the above notation) it is reasonable to consider that the binding energy is equal to the permeation energy, i.e. ~2.0 eV [15]. Under this assumption, one can assign the high temperature TDS peak (usually seen in low flux exposures around 700-900 K) to the release from bubbles [16]. However, the TDS spectra obtained after the high flux exposures, at high temperature (above 600 K) and/or high dose (above 10^{26} D/m^2), reveal broadening or presence of extra peaks above 900 K [13, 17]. These high temperature release stages cannot be simply explained by detrapping from voids with the binding energy of 2.0 eV, while there are no reasons to assume the presence of any stronger traps than voids/cavities.

To address the issue of high temperature detrapping stage, we analyse the information available from atomic scale. *Ab initio* and Molecular Dynamics simulations showed that H tends to occupy off-centred position in a vacancy, experiencing weak attraction to W atoms [18, 19]. This implies that H atoms, when filling a void, first should occupy the inner surface positions and then fill the centre.

Thus for the binding energy corresponding to the H trapped at nm size bubbles (type IV defects) we used the results of *ab initio* calculations to account for the transition of H atom from adsorbed state on W surface into the bulk material [15]. These calculations reported 1.96 eV as a value of the binding energy and attempt frequency of 1.07×10^{13} s⁻¹. However, the expression for the release rate should be different from Eq. 1, since only atoms in adsorbed state on the inner surface of the bubble are available for the detrapping reaction. Thus, the release from nm-scale bubbles is defined by:

$$R_B^- = C_T N_S \nu_{ads} \exp(-\frac{E_b^{ads} + E_m}{kT}) (2)$$

where C_T – trap concentration, N_S – number of available sites for the reaction, we estimate this as a ratio between an inner surface area of the bubble and unit surface for the reaction reported in [15] $(\sqrt{2}a_0^2, a_0 - \text{lattice unit}), V_{ads}, E_b^{ads}$ attempt frequency and binding energy from [15]. To create a universal binding energy master curve accounting for a size of the trapping defect (i.e. number of vacancies released), we propose the following expression:

$$R^{-} = R_{B}^{-} + (R_{J}^{-} - R_{B}^{-})N_{V}^{-1/3} (3)$$

here R_B^- and R_J^- are the release rates for bubble and jog correspondingly, N_V – number of empty lattice sites in tungsten lattice which the trapping defect occupies. The illustration of the binding energy variation as a function of N_V is presented in Fig. 1. To compute the H release profile from a defect (of a given size and for a given temperature ramp), we assume that after the exposure the defect is filled with H up to a critical pressure (6.5 H atoms per W site), following the analysis done in [11] based on the jog-punching mechanism. Then, the H release rate from the defect is calculated as a function of increasing temperature. The goal of such calculations is to define the position of the release peak, depending on the size of the defect and concentration of H inside.



Pic. 1. Illustration of binding energy master curve used in simulations. R – defect open volume radius, N_V – corresponding number of W lattice sites that defect occupies.

3. Results and discussion

Following the above formulated equations, we analyse the TDS spectra obtained in [13]. In Fig. 2, a comparison of three normalized TDS spectra with the release rate according to the calculations is represented. Normalization of the TDS spectra was done by dividing each one by the maximum release rate value, since here we are interested in correspondence of the peak positions, but not the absolute values of H release. As can be seen form Fig. 2, TDS spectra for the low flux (fig. b) and high flux high temperature exposure (fig. a) reveals similar single peak as in the spectrum measured after ~870 K exposure. Temperature position of the peak corresponds to the release from 5-10 nm bubbles, following our calculations. TDS spectrum for high flux low temperature exposure (blue triangles) shows two peaks at lower temperature, whose positions can be well fit by imposing the release rate from 1.5 nm and 0.5 nm defects (blue and green lines in Fig. 2a).



Fig 2. Comparison of normalized TDS spectra from work [13] with results of simulation for high flux $(5x10^{23} \text{ D/m}^2 \text{s} \text{ at } 530 \text{ K} \text{ and } 870 \text{ K})$ (a) and low flux $(9x10^{21} \text{ D/m}^2 \text{s} \text{ at } 530 \text{ K} \text{ and } 630 \text{ K})$ (b) exposures.

The difference in shape and peak position of the TDS spectra can be related to a size of releasing defects, and in turn interlinked with the trapping mechanism, governing H retention depending on exposure conditions. In [11], the prediction of experimental flux – temperature combination for the dislocation-driven H trapping leading to the formation of super-critical H clusters (contributing to the TDS release at high temperature) was made. The flux-temperature combination favouring the nucleation of super-critical H bubbles is represented by a grey area in Fig.3. As one can see, only high flux low temperature exposure is within the range of the dislocation-mediated retention. At the same time, all three other dots are outside the defined area. This is consistent with the fact that only the TDS spectrum after high flux low temperature exposure exhibits two peaks (low temperature peak can be attributed to1 nm size defects). Whereas, all other spectra show only one peak corresponding to bubbles of 5-10 nm size. Small (1 nm scale) defects can be seen as traces of bubble nucleation on dislocation and their subsequent growth. While, if only large defects contribute to the TDS spectra, it is very likely that other mechanisms (such as triple junctions, natural porosity, etc.) but dislocation-driven trapping govern H retention and bubble formation.



Fig. 3. Comparison of experimental conditions from work [13] with prediction of the model for conditions for dislocation mediated retention.

4. Conclusive remarks

The proposed modification for H release rate from nano-metric pressurized bubbles allows one to relate positions of TDS peaks and explain a shift of high-temperature peak up to 200 K under exposure conditions favoring the growth of bubbles. The occurrence of a single-peak or multiple-peak in TDS spectra can be related to the mechanism of H bubble formation, depending on plasma flux and surface temperature. The proposed here computational analysis was applied to experimental data from [13]. According to it, only a combination of high flux and low temperature exposure $(530 \text{ K} - 5 \times 10^{23} \text{ D/m}^2)$ promotes the nucleation and growth of H bubbles on dislocations, as proposed earlier by jog punching mechanism [11]. In these exposure conditions, the TDS spectra exhibit two well defined peaks attributed to 1 nm size H clusters (presumably, H is trapped in jogs on dislocation lines) and H bubbles (resulting in the high temperature release peak). The three other TDS spectra exhibit only one peak (at high temperature) to be attributed to H bubbles with a size of 5-10 nm, nucleated at other "natural" traps such as random grain boundaries and their junctions. Validation of the dislocation-driven mechanism and proposed here H release model can be realized by performing dedicated exposures in 450-600 K temperature range at flux of 10^{23} D/m²/s, which should result in the dislocation-driven trapping below ~500 K to be seen in TDS spectrum as a peak conventionally attributed to vacancy-like defects (i.e. positioned around 600-700 K).

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