

Open Access

Synthesis of 2-amino-3-arylpropan-1-ols and 1-(2,3-diaminopropyl)-1,2,3-triazoles and evaluation of their antimalarial activity

Matthias D'hooghe^{*1}, Stéphanie Vandekerckhove¹, Karen Mollet^{1,§}, Karel Vervisch¹, Stijn Dekeukeleire^{1,§}, Liesbeth Lehoucq¹, Carmen Lategan², Peter J. Smith², Kelly Chibale³ and Norbert De Kimpe^{*1}

doi:10.3762/bjoc.7.205

Full Research Paper

Address:

¹Department of Sustainable Organic Chemistry and Technology, Faculty of Bioscience Engineering, Ghent University, Coupure Links 653, B-9000 Ghent, Belgium, ²Medical School, University of Cape Town, K45, OMB, Groote Schuur Hospital, Observatory, 7925, South Africa and ³Department of Chemistry and Institute of Infectious Disease & Molecular Medicine, University of Cape Town, Rondebosch 7701, South Africa

Email:

Matthias D'hooghe^{*} - matthias.dhooghe@UGent.be; Norbert De Kimpe^{*} - norbert.dekimpe@UGent.be

* Corresponding author § Aspirant of the "Research Foundation – Flanders" (FWO-Vlaanderen)

Keywords:

aminopropanes; antimalarial activity; aziridines; β -lactams; ring opening

Received: 28 September 2011 Accepted: 28 November 2011 Published: 30 December 2011 Associate Editor: J. N. Johnston © 2011 D'hooghe et al; licensee Beilstein-Institut. License and terms: see end of document.

Beilstein J. Org. Chem. 2011, 7, 1745-1752.

Abstract

A variety of 2-amino-3-arylpropan-1-ols, *anti*-2-amino-3-aryl-3-methoxypropan-1-ols and *anti*-2-amino-1-arylpropan-1,3-diols were prepared selectively through elaboration of *trans*-4-aryl-3-chloro- β -lactams. In addition, a number of 2-(azidomethyl)-aziridines was converted into novel 2-[(1,2,3-triazol-1-yl)methyl]aziridines by Cu(I)-catalyzed azide-alkyne cycloaddition, followed by microwave-assisted, regioselective ring opening by dialkylamine towards 1-(2,3-diaminopropyl)-1,2,3-triazoles. Although most of these compounds exhibited weak antiplasmodial activity, six representatives showed moderate antiplasmodial activity against both a chloroquine-sensitive and a chloroquine-resistant strain of *Plasmodium falciparum* with IC₅₀-values of $\leq 25 \mu$ M.

Introduction

Malaria remains a major issue in health control, especially in developing countries. This disease affects 40% of the global population, causing an annual mortality of one million people [1]. Despite recent advances in the development of a vaccine against malaria, chemotherapy remains the most viable alternative towards treatment of the disease [2]. In light of the rapid

Beilstein J. Org. Chem. 2011, 7, 1745-1752.

emergence of multiple drug resistance to clinically established antimalarial drugs, however, there is a compelling need to introduce new chemicals that can overcome this resistance. In 2007, nitrogen-analogues of glycerol, which have a long-standing tradition in medicine as β -blockers, were introduced as a novel class of antimalarials [3]. Prior to this, the well known β-blocker propranolol was shown to inhibit infection of erythrocytes by P. falciparum, as well as to reduce the parasitaemia of P. berghei infections in vivo [4,5]. In light of the biological potential of these compounds, continuous efforts have been devoted to the preparation of structurally diverse analogues bearing a functionalized propane skeleton [6-8]. In that respect, we have been engaged in the stereoselective synthesis of syn-2alkoxy-3-amino-3-arylpropan-1-ols 1 by reductive ring opening of the corresponding β -lactams, which were shown to be of great importance as novel antiplasmodial agents (Figure 1) [6]. More recently, we reported 1,2,3-triaminopropanes 2 as a new class of antimalarial compounds (Figure 1), prepared through microwave-assisted, regioselective ring opening of the corresponding 2-(aminomethyl)aziridines by diethylamine [8]. Nonetheless, a number of challenges with regard to structure-activity relationship studies of functionalized aminopropanes remain unaddressed, especially concerning the screening of structural analogues of aminopropanes 1 and 2.

In the present paper, the synthesis of racemic *anti*-2-aminopropan-1-ols **3** is described as a variant of the *syn*-3-aminopropan-1-ols **1** synthesis (regio- and stereoisomerism with respect to the relative position of the amino group NHR² and the oxygen substituent OR³, Figure 1), by applying a different synthetic route. Furthermore, a new synthetic approach is disclosed towards racemic aminopropanes **4** bearing a 1,2,3triazole moiety, as structural analogues of the previously reported 1,2,4-triazoles **2** (Figure 1). Both classes of functionalized aminopropanes **3** and **4** were tested for their antiplasmodial activity.

Results and Discussion Synthesis

Within azaheterocyclic chemistry, aziridines [9-17] and β -lactams [18-27] are extraordinary classes of strained com-

pounds with diverse synthetic and biological applications. In previous works, we have elaborated the synthetic potential of 3-chloroazetidin-2-ones with a focus on stereoselectivity, thus providing convenient entries into, e.g., aziridines, azetidines and β-aminoalcohols [28-31]. In continuation of our interest in the use of functionalized β-lactams as synthons for further elaboration, racemic *trans*-4-aryl-3-chloro-β-lactams 5 were prepared by treatment of N-(arylmethylidene)alkylamines (synthesized in high yields through condensation of the corresponding benzaldehydes with the appropriate primary amines in CH₂Cl₂ in the presence of anhydrous MgSO₄) with 1.5 equiv of chloroacetyl chloride and 3 equiv of 2,6-lutidine in benzene according to a literature protocol [30]. Subsequently, β -lactams 5 were subjected to LiAlH₄-mediated reductive ring opening, furnishing either 2-aminopropan-1-ols **6a–c**, by using two molar equiv of LiAlH₄ in Et₂O under reflux for 20-80 h, or trans-2aryl-3-(hydroxymethyl)aziridines 7a-h by applying milder reactions conditions (i.e., one molar equiv of LiAlH₄ in Et₂O at room temperature for 5-8 h) (Scheme 1) [30].

As aziridines are known to be versatile synthetic intermediates for the preparation of a variety of ring-opened and ringexpanded amines, the aziridines 7 were deployed as substrates for the stereoselective synthesis of functionalized aminopropanols. In accordance with a literature approach [30], the nonactivated *trans*-2-aryl-3-(hydroxymethyl)aziridines 7 were regio- and stereoselectively converted into anti-2-amino-3-aryl-3-methoxypropan-1-ols 8a-e through heating in methanol under reflux (Scheme 1). Furthermore, in order to provide access to the class of 2-aminopropan-1,3-diols, aziridines 7 were evaluated for the first time as substrates for a water-induced aziridine ring opening in an acidic medium. Thus, treatment of trans-2-aryl-3-(hydroxymethyl)aziridines 7 with one equiv of para-toluenesulfonic acid in a H₂O/THF (1/1) solvent system [32] furnished novel anti-2-amino-1-arylpropan-1,3-diols 9a-d in good yields after 3 h at 40 °C, again in a regio- and stereospecific way (Scheme 1). The observed regio- and stereoselectivity in aminopropanols 8 and 9 can be rationalized by considering the ring opening of the aziridine moiety at C2 due to benzylic stabilization of the developing carbenium ion in an S_N2 fashion [30].





As the use of imines bearing a N-tert-butyl group in combination with a substituent in the ortho-position of the aromatic ring is known to afford the corresponding cis-4-aryl-3-chloroazetidin-2-ones as the major stereoisomers after condensation with chloroketene in benzene [30], racemic cis-3-chloro-βlactams 10a,b were prepared and converted into cis-2-aryl-3-(hydroxymethyl)aziridines 11a,b upon treatment with two molar equiv of LiAlH₄ in Et₂O under reflux for 15 h (Scheme 2). Next, the aziridines 11, which have previously been shown to be unreactive towards LiAlH₄ and methanol and thus unable to undergo ring opening [30], were used as substrates for a water-induced ring opening through initial protonation of the aziridine ring with p-TsOH. Although more drastic reaction conditions were required compared to the ring opening of trans-2-aryl-3-(hydroxymethyl)aziridines 7 (3 equiv p-TsOH, Δ , 30 h instead of 1 equiv p-TsOH, 40 °C, 3 h), novel syn-aminopropanols 12a,b were obtained in a selective and convenient way (Scheme 2, yields after purification), providing the first example of the ring opening of this type of aziridines. Also

in this case, the observed regio- and stereoselectivity in the formation of aminopropanols 12 can be rationalized by considering the ring opening of the aziridine moiety at C2 due to benzylic stabilization of the developing carbenium ion in an S_N2 fashion [30]. The formation of the other regio- and stereoisomers was excluded based on detailed spectroscopic analysis.

It should be noted that both diastereomeric antipodes of the class of 1-aryl-2-aminopropan-1,3-diols, i.e., *anti-* and *syn-* aminopropanols **9** and **12**, can now be prepared selectively through choice of the appropriate imine for the Staudinger synthesis of the starting β -lactams.

Given the recently disclosed antiplasmodial activities of a number of 2,3-diamino-1-(1,2,4-triazol-1-yl)propanes [8], the second objective of this work was the preparation of new analogues bearing a 1,2,3-triazole moiety instead. A powerful methodology towards the synthesis of functionalized 1,2,3-tri-



azoles involves the Cu(I)-catalyzed azide-alkyne Huisgen cycloaddition (CuAAC) [33], which has gained major interest from the synthetic community due to its high efficiency and selectivity. Eligible substrates to perform "click chemistry" [34] incorporate an azide group and an aziridine ring in their structure, for example in 2-(azidomethyl)aziridines, thus providing a direct access to 2-[(1,2,3-triazol-1-yl)methyl]aziridines through Cu-catalyzed reaction with alkynes [35].

In this work, nonactivated N-(arylmethyl)aziridines were selected as substrates, as previous research had revealed the importance of N-benzyl, N-chlorobenzyl and N-methoxybenzyl groups in functionalized aminopropanes with regard to their antiplasmodial activity [6,8]. Thus, a number of racemic 2-(azidomethyl)aziridines 14 was prepared by reaction of sodium azide with 2-(bromomethyl)aziridines 13 [36-38], employing the electrophilicity of the latter as a convenient handle for their connection to other moieties. In this way, novel 2-(azidomethyl)aziridines 14a-e were prepared in good yields by treatment of bromides 13 with two equiv of sodium azide in DMSO at 80 °C for 16 h (CAUTION) (Scheme 3). Subsequently, a CuI-catalyzed 1,3-cycloaddition of N-(arylmethyl)aziridine azides 14 was evaluated for the first time by utilizing one equiv of an arylacetylene in CH₃CN under reflux for 16 h, furnishing a direct entry towards new 2-[(1,2,3-triazol-1yl)methyl]aziridines 15a-f in a highly efficient and selective way (Scheme 3).

The final step comprised ring opening of the aziridine moiety in compounds **15** by diethylamine to afford functionalized aminopropanes as potential antimalarial agents. With the intention to introduce a diethylamino group, a microwave-promoted ring opening of analogous aziridines by diethylamine with the aid of the Et₂NH·HCl/Et₂NH system was employed [8]. In this way, protonation of the aziridine ring provides a highly electrophilic aziridinium intermediate, which is prone to undergo nucleophilic ring opening. In order to drive the reaction to completion, and to avoid competition between chloride- and diethylamine-induced ring opening, a large excess of Et₂NH·HCl (20 equiv) and an additional amount of diethylamine (10 equiv) was used. Thus, heating of the aziridines 15 at 140 °C in CH₃CN under microwave irradiation resulted in full and selective conversion to the desired new triaminopropanes 16a-e after 2 h (Scheme 3), which were purified by column chromatography (SiO₂) in order to obtain analytically pure samples. Furthermore, in addition to the use of diethylamine, the introduction of a dimethylamino group was considered in order to compare the contribution of this moiety to the potential antiplasmodial activity with that of a diethylamino group. This objective was achieved by treatment of aziridines 15 with 20 equiv of Me₂NH·HCl in CH₃CN at 140 °C for 2 h under microwave irradiation, resulting in dimethylaminopropanes 16f,g in good yields (Scheme 3). This result showed that this microwave-assisted methodology for the ring opening of nonactivated aziridines can be further extended towards the use of other secondary amines. In order to introduce structural diversity within these molecules, different substituent patterns at the aromatic rings (R^1, R^2) in aminopropanes **16a–g** were realized as well.

In view of the biological potential of aminopropanes in general, compounds 6, 8, 9, 12 and 16 were subsequently screened for their antiplasmodial activity.

In addition, aziridines **14** and **15** were tested against the malaria parasite *Plasmodium falciparum*, too.



Biological evaluation

At first, compounds **6a–c**, **8a–e**, **9a–d**, **12a,b**, **14a–e**, **15a–f** and **16a–g** were screened for in vitro antiplasmodial activity. All samples were tested in triplicate on one occasion against a chloroquine-sensitive (CQS) strain of *P. falciparum* (D10). Those samples showing antiplasmodial activity were then tested in triplicate on one occasion against a chloroquine-resistant (CQR) strain of *P. falciparum* (Dd2) and screened for in vitro cytotoxicity against a Chinese hamster ovary (CHO) cell-line,

in triplicate on one occasion. The antiplasmodial and cytotoxicity assays were performed as described previously [8,39,40].

The results from the biological study are summarized in Table 1. Although most of these compounds were shown to possess weak or no antiplasmodial activity, eight of them (i.e., compounds **6a**, **9a**,**b**, **15c**,**d**, **16a**,**b**,**f**) were identified as potentially interesting for further study with IC₅₀-values of \leq 25 µM. Moreover, these compounds, with the exception of triamino-

Compound	R ¹	R ²	R ³	D10: IC ₅₀ (µM)	Dd2: IC ₅₀ (μΜ)	CHO: IC ₅₀ (µM)	Rl ^a	SIb
6a	4-Cl	Bn	_	12.58	10.88	137.90	0.9	11
6b	4-OMe	iBu	-	281.58	ND	ND	ND	ND
6c	2-F	<i>n</i> -Pr	-	217.34	ND	ND	ND	ND
8a	Н	iPr	-	369.70	ND	ND	ND	ND
8b	Н	Bn	-	143.79	ND	ND	ND	ND
8c	4-Cl	Bn	-	198.56	ND	ND	ND	ND
8d	4-OMe	iBu	_	38.56	25.69	>530	0.7	ND
8e	2-F	<i>n</i> -Pr	-	38.54	21.92	>530	0.6	ND
9a	4-Cl	Bn	_	25.22	8.47	>530	0.3	ND
9b	2-CI	Bn	_	21.18	13.57	>530	0.6	ND
9c	4-Me	Bn	_	129.86	ND	ND	ND	ND
9d	Н	4-CIBn	_	80.68	ND	ND	ND	ND
12a	OMe	_	_	98.68	ND	ND	ND	ND
12b	F	_	_	60.46	ND	ND	ND	ND
14a	Н	_	_	230.40	466.81	>530	2	ND
14b	4-Cl	_	_	100.82	ND	ND	ND	ND
14c	4-OMe	_	_	160.50	ND	ND	ND	ND
14d	2-CI	_	_	93.68	ND	ND	ND	ND
14e	3-CI	_	_	179.86	ND	ND	ND	ND
15a	H	Н	_	34.44	106.97	>530	3.1	ND
15b	4-Cl	Н	_	41.32	ND	ND	ND	ND
15c	4-OMe	Н	_	25.69	55.65	>530	2.2	ND
15d	Н	Me	_	20.43	20.99	>530	1	ND
15e	Н	OMe	_	40.54	ND	ND	ND	ND
15f	2-Cl	H	_	32.33	25.00	>530	0.8	ND
16a	H	Н	Et	22.09	231.47	>530	10.5	ND
16b	4-Cl	Н	Et	25.86	176.40	>530	6.8	ND
16c	4-OMe	Н	Et	139.61	ND	ND	ND	ND
16d	Н	Me	Et	32.55	ND	ND	ND	ND
16e	н	OMe	Et	171.37	ND	ND	ND	ND
166 16f	2-Cl	H	Me	11.33	13.03	181.83	1.2	16.
16g	H	н	Me	69.58	ND	ND	ND	ND
CQ				19.14 ng/mL (<i>n</i> = 6)	75.56 ng/mL (<i>n</i> = 5)		3.9	
Emetine						0.27 (<i>n</i> = 6)		

^aRI (Resistance Index) = IC_{50} Dd2/ IC_{50} D10; ^bSI (Selectivity Index) = IC_{50} CHO/ IC_{50} D10; ND = not determined; *n* = number of data sets averaged. The more hydrophobic samples were added to the parasites as a suspension, meaning that for these samples the reported IC_{50} -value might be an underestimation of the activity.

propanes **16a,b**, also proved to be active against a chloroquineresistant strain of *P. falciparum* (Dd2). In addition, the in vitro cytotoxicity results showed that only compounds **6a** and **16f** have lower selectivity with SI's of 11 and 16, respectively, whereas the other compounds did not show cytotoxicity at the concentrations tested.

Although the aziridine moiety was initially only considered as a synthetically useful entity, two 2-[(1,2,3-triazol-1-yl)methyl]aziridines (15c and 15d) were also found to exhibit weak antiplasmodial activity. On the other hand, the aminopropane unit has again proven its value as a template for the preparation of novel antimalarial agents, as a variety of structurally different aminopropanes were demonstrated to exhibit weak to moderate antiplasmodial activity. In particular, antiplasmodial assays against a chloroquine-sensitive strain of P. falciparum (D10) showed activity for 2-aminopropan-1-ol 6a, 2-aminopropan-1,3-diols 9a,b and triaminopropanes 16a,b,f with IC₅₀-values between 11.3 and 25.9 µM. Moreover, screening against a chloroquine-resistant strain of P. falciparum (Dd2) revealed antiplasmodial activity for 2-aminopropan-1-ol 6a, 2-aminopropan-1,3-diols 9a,b and triaminopropane 16f with IC₅₀values between 8.5 and 13.6 μ M.

From a structure–activity relationship viewpoint, the presence of a chlorinated aromatic ring seems to contribute to the antiplasmodial activity of these functionalized aminopropanes, and the introduction of a dimethylamino group at the expense of a diethylamino moiety might provide better activities in some cases. It is noteworthy that these compounds were synthesized in racemic form, and it is conceivable that enantiomerically pure variants could deliver superior activities. It should also be noted that, in general, the bioactivities reported in this paper are less pronounced as compared to those described in literature precedents on the synthesis and evaluation of functionalized aminopropanes [6,8].

Conclusion

In summary, a variety of 2-amino-3-arylpropan-1-ols, *anti*-2amino-3-aryl-3-methoxypropan-1-ols and *anti*-2-amino-1-arylpropan-1,3-diols were prepared selectively through elaboration of *trans*-4-aryl-3-chloro- β -lactams. Furthermore, a number of 2-(azidomethyl)aziridines were converted into novel 2-[(1,2,3triazol-1-yl)methyl]aziridines by Cu(I)-catalyzed azide-alkyne cycloaddition, followed by microwave-assisted, regioselective ring opening by diethyl- or dimethylamine towards the corresponding 1-(2,3-diaminopropyl)-1,2,3-triazoles. From a synthetic viewpoint, new insights were provided concerning the water-induced ring opening of nonactivated *cis*- and *trans*-2aryl-3-(hydroxymethyl)aziridines and with respect to the synthesis and use of 1-arylmethyl-2-(azidomethyl)aziridines for azide-alkyne cycloaddition reactions. From a biological viewpoint, most of these compounds exhibited weak antiplasmodial activity, although six representatives showed moderate antiplasmodial activity against both a chloroquine-sensitive and a chloroquine-resistant strain of *P. falciparum* with IC₅₀-values of \leq 25 µM.

Experimental

General information regarding NMR, IR, MS and elemental analyses, melting point measurements, and microwave reaction conditions can be found in the literature [8].

anti-2-(*N*-Benzylamino)-1-(4-chlorophenyl)propan-1,3-diol (**9a**)

Recrystallization from hexane/EtOAc (1:25), white crystals, 95%. Mp 192.3 °C; ¹H NMR (300 MHz, CDCl₃) δ 2.39 (s, 2H), 3.12–3.18 (m, 1H), 3.54 (dd, *J* = 13.0, 3.6 Hz, 1H), 3.82 (dd, *J* = 13.0, 5.8 Hz, 1H), 4.27 and 4.32 (2 d, *J* = 13.2 Hz, 2 × 1H), 5.24 (d, *J* = 2.2 Hz, 1H), 7.04–7.07, 7.19–7.36, 7.46–7.49 and 7.69–7.71 (4 m, 2H, 3H, 2H, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 51.0, 59.4, 62.6, 72.0, 127.2, 128.1, 128.7, 128.8, 128.9, 133.4, 137.0, 139.1; IR (cm⁻¹) v_{max}: 3437 (OH), 3342 (NH), 2926, 1448, 1162, 1008, 681; MS (70 eV) *m/z* (%): 292/4 (M⁺ + 1, 100); HRMS (ESI): [M + H]⁺ calcd for C₁₆H₁₉ClNO₂, 292.1104; found, 292.1112.

syn-2-(*N-tert*-Butylamino)-1-(2-methoxyphenyl)propan-1,3-diol (**12a**)

*R*_f 0.07 (EtOAc), white crystals, 30%. Mp 112.8 °C; ¹H NMR (300 MHz, CDCl₃) δ 1.03 (s, 9H), 2.77 (ddd, *J* = 7.1, 3.0, 3.0 Hz, 1H), 3.50–3.51 (m, 2H), 3.86 (s, 3H), 4.80 (d, *J* = 7.1 Hz, 1H), 6.87–6.89, 6.98–7.03, 7.22–7.28 and 7.47–7.50 (4 m, 4 × 1H); ¹³C NMR (75 MHz, CDCl₃) δ 30.0, 50.8, 55.6, 58.1, 63.7, 68.5, 110.4, 121.3, 127.7, 128.4, 130.5, 156.5; IR (cm⁻¹) v_{max}: 3391 (NH), 3313 (OH), 3058, 3004, 2987, 2957, 2927, 2873, 2838, 1492, 1467, 1438, 1368, 1242, 1066, 1050, 1027, 755; MS (70 eV) *m/z* (%): 254 (M⁺ + 1, 100); HRMS (ESI): [M + H]⁺calcd for C₁₄H₂₄NO₃, 254.1756; found, 254.1766.

2-Azidomethyl-1-(phenylmethyl)aziridine (14a)

 $R_{\rm f}$ 0.20 (hexane/EtOAc 4:1), yellow oil, 70%. ¹H NMR (300 MHz, CDCl₃) δ 1.51 (d, J = 6.1 Hz, 1H), 1.79 (d, J = 3.3 Hz, 1H), 1.79–1.86 (m, 1H), 3.18 and 3.27 (2 dd, J = 12.9, 6.6, 4.4 Hz, 2H) 3.38 and 3.58 (2 d, J = 13.2 Hz, 2H), 7.24–7.36 (m, 5H); ¹³C NMR (75 MHz, CDCl₃) δ 32.2, 37.9, 53.8, 64.3, 127.4, 128.3, 128.6, 138.8; IR (cm⁻¹) v_{max}: 2088 (N₃), 1453, 1357, 1322, 1255, 1159, 1062, 1028, 907, 732, 697; MS (70 eV) m/z (%): 189 (M⁺ + 1, 100); HRMS (ESI): [M + H]⁺ calcd for C₁₀H₁₃N₄, 189.1140; found, 189.1141.

1-Phenylmethyl-2-[(4-phenyl-1,2,3-triazol-1yl)methyl]aziridine (**15a**)

*R*_f 0.23 (CHCl₃/MeOH 98:2), viscous light-brown oil, 61%. ¹H NMR (300 MHz, CDCl₃) δ 1.65 (d, *J* = 6.6 Hz, 1H), 1.88 (d, *J* = 3.3 Hz, 1H), 2.01–2.08 (m, 1H), 3.09 and 3.67 (2 d, *J* = 12.9 Hz, 2H), 3.93 and 4.70 (2 dd, *J* = 14.3, 8.2, 3.3 Hz, 2H), 7.06–7.43 (m, 8H), 7.49 (s, 1H), 7.66–7.69 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 32.4, 38.3, 53.5, 64.5, 119.9, 125.9, 127.7, 128.1, 128.4, 128.6, 128.7, 130.8, 138.2, 147.9; IR (cm⁻¹) v_{max}: 2919, 1453, 1358, 1225, 1075, 1046, 1027, 763, 731, 694; MS (70 eV) *m/z* (%): 291 (M⁺ + 1, 100); HRMS (ESI): [M + H]⁺ calcd for C₁₈H₁₉N₄, 291.1610; found, 291.1613.

3-Diethylamino-2-(phenylmethyl)amino-1-(4phenyl-1,2,3-triazol-1-yl)propane (**16a**)

*R*_f 0.19 (CHCl₃/MeOH 97:3), light-brown oil, 56%. ¹H NMR (300 MHz, CDCl₃) δ 0.92 (t, *J* = 7.2 Hz, 6H), 2.26–2.52 (m, 6H), 3.05–3.13 (m, 1H), 3.74 and 3.81 (2 d, *J* = 13.5 Hz, 2H), 4.36 and 4.44 (2 dd, *J* = 14.2, 4.9, 4.7 Hz, 2H), 7.21–7.44 and 7.82–7.84 (2 m, 10H), 7.86 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 11.8, 47.1, 52.0, 52.3, 55.0, 55.4, 121.2, 125.8, 127.2, 128.1, 128.2, 128.6, 128.9, 130.9, 140.2, 147.5; IR (cm⁻¹) v_{max}: 2967, 1462, 1454, 1073, 762, 734, 694; MS (70 eV) *m/z* (%): 364 (M⁺ + 1, 100); HRMS (ESI): [M + H]⁺ calcd for C₂₂H₃₀N₅, 364.2501; found, 364.2503.

Supporting Information

Supporting Information File 1

Experimental procedures and characterization data. [http://www.beilstein-journals.org/bjoc/content/ supplementary/1860-5397-7-205-S1.pdf]

Acknowledgements

The authors are indebted to the "Research Foundation – Flanders" (FWO-Vlaanderen), the "Institute for the Promotion of Innovation through Science and Technology – Flanders" (IWT-Vlaanderen), and to Ghent University (GOA) for financial support.

References

- 1. WHO Fact sheet on malaria N°94, January 2009.
- Gemma, S.; Travagli, V.; Savini, L.; Novellino, E.; Campiani, G.; Butini, S. *Recent Pat. Antiinfect. Drug. Discov.* 2010, *5*, 195.
- Robin, A.; Brown, F.; Bahamontes-Rosa, N.; Wu, B.; Beitz, E.; Kun, J. F. J.; Flitsch, S. L. *J. Med. Chem.* 2007, *50*, 4243. doi:10.1021/jm070553l
- Murphy, S. C.; Harrison, T.; Hamm, H. E.; Lomasney, J. W.; Mohandas, N.; Haldar, K. *PLoS Med.* 2006, *3*, 2403. doi:10.1371/journal.pmed.0030528

- Harrison, T.; Samuel, B. U.; Akompong, T.; Hamm, H.; Mohandas, N.; Lomasney, J. W.; Haldar, K. *Science* 2003, *301*, 1734. doi:10.1126/science.1089324
- D'hooghe, M.; Dekeukeleire, S.; Mollet, K.; Lategan, C.; Smith, P. J.; Chibale, K.; De Kimpe, N. *J. Med. Chem.* **2009**, *52*, 4058. doi:10.1021/jm9002632
- Pérez-Silanes, S.; Berrade, L.; García-Sánchez, R. N.; Mendoza, A.; Galiano, S.; Pérez-Solórzano, B. M.; Nogal-Ruiz, J. J.; Martínez-Fernández, A. R.; Aldana, I.; Monge, A. *Molecules* 2009, *14*, 4120. doi:10.3390/molecules14104120
- D'hooghe, M.; Kenis, S.; Vervisch, K.; Lategan, C.; Smith, P. J.; Chibale, K.; De Kimpe, N. *Eur. J. Med. Chem.* **2011**, *46*, 579. doi:10.1016/j.ejmech.2010.11.037
- Zwanenburg, B.; ten Holte, P. *Top. Curr. Chem.* 2001, *216*, 93. doi:10.1007/3-540-44726-1_3
- 10. Sweeney, J. B. Chem. Soc. Rev. 2002, 31, 247. doi:10.1039/b006015
- 11. Hu, X. E. Tetrahedron 2004, 60, 2701. doi:10.1016/j.tet.2004.01.042
- 12. Tanner, D. Angew. Chem., Int. Ed. Engl. **1994**, 33, 599. doi:10.1002/anie.199405991
- Osborn, H. M. I.; Sweeney, J. Tetrahedron: Asymmetry 1997, 8, 1693. doi:10.1016/S0957-4166(97)00177-8
- 14. McCoull, W. M.; Davis, F. A. *Synthesis* **2000**, 1347. doi:10.1055/s-2000-7097
- Watson, I. D. G.; Yu, L.; Yudin, A. K. Acc. Chem. Res. 2006, 39, 194. doi:10.1021/ar050038m
- 16. Padwa, A.; Murphree, S. S. ARKIVOC 2006, No. iii, 6.
- Singh, G. S.; D'hooghe, M.; De Kimpe, N. Chem. Rev. 2007, 107, 2080. doi:10.1021/cr0680033
- Ojima, I.; Delaloge, F. Chem. Soc. Rev. 1997, 26, 377. doi:10.1039/cs9972600377
- Alcaide, B.; Almendros, P. Synlett 2002, 381. doi:10.1055/s-2002-20448
- Fisher, J. F.; Meroueh, S. O.; Mobashery, S. Chem. Rev. 2005, 105, 395. doi:10.1021/cr030102i
- 21. Alcaide, B.; Almendros, P.; Aragoncillo, C. Chem. Rev. 2007, 107, 4437. doi:10.1021/cr0307300
- 22. Ojima, I. Acc. Chem. Res. 1995, 28, 383. doi:10.1021/ar00057a004
- 23. Singh, G. S. *Tetrahedron* **2003**, *59*, 7631. doi:10.1016/S0040-4020(03)01099-8
- 24. Singh, G. S.; D'hooghe, M.; De Kimpe, N. In Comprehensive Heterocyclic Chemistry III; Katritzky, A.; Ramsden, C.; Scriven, E.; Taylor, R., Eds.; Elsevier: Oxford, 2008; Vol. 2, p 1. doi:10.1016/B978-008044992-0.00201-7
- 25. Xing, B.; Rao, J.; Liu, R. Mini-Rev. Med. Chem. 2008, 8, 455. doi:10.2174/138955708784223558
- 26. Aranda, M. T.; Perez-Faginas, P.; Gonzalez-Muniz, R. *Curr. Org. Synth.* **2009**, *6*, 325. doi:10.2174/157017909788921899
- D'hooghe, M.; Dekeukeleire, S.; Leemans, E.; De Kimpe, N. Pure Appl. Chem. 2010, 82, 1749. doi:10.1351/PAC-CON-09-09-39
- Van Driessche, B.; Van Brabandt, W.; D'hooghe, M.; Dejaegher, Y.; De Kimpe, N. *Tetrahedron* 2006, *62*, 6882.
- doi:10.1016/j.tet.2006.04.104
 29. D'hooghe, M.; Dekeukeleire, S.; De Kimpe, N. Org. Biomol. Chem.
 2008, 6, 1190. doi:10.1039/b719686e
- D'hooghe, M.; Mollet, K.; Dekeukeleire, S.; De Kimpe, N. Org. Biomol. Chem. 2010, 8, 607. doi:10.1039/b919864d
- Mollet, K.; D'hooghe, M.; De Kimpe, N. J. Org. Chem. 2011, 76, 264. doi:10.1021/jo1020932

- Manaka, T.; Nagayama, S.-I.; Desadee, W.; Yajima, N.; Kumamoto, T.; Watanabe, T.; Ishikawa, T.; Kawahata, M.; Yamaguchi, K. *Helv. Chim. Acta* 2007, *90*, 128. doi:10.1002/hlca.200790006
- 33. Hein, J. E.; Fokin, V. V. Chem. Soc. Rev. 2010, 39, 1302. doi:10.1039/b904091a
- 34. Kolb, H. C.; Finn, M. G.; Sharpless, K. B. Angew. Chem., Int. Ed. 2001, 40, 2004.

doi:10.1002/1521-3773(20010601)40:11<2004::AID-ANIE2004>3.0.CO;2-5

- Jamookeeah, C. E.; Beadle, C. D.; Harrity, J. P. A. Synthesis 2009, 133. doi:10.1055/s-0028-1083270
- 36. De Kimpe, N.; Jolie, R.; De Smaele, D. J. Chem. Soc., Chem. Commun. 1994, 1221. doi:10.1039/C39940001221
- D'hooghe, M.; Waterinckx, A.; Kimpe, N. J. Org. Chem. 2005, 70, 227. doi:10.1021/jo048486f
- D'hooghe, M.; Rottiers, M.; Jolie, R.; De Kimpe, N. Synlett 2005, 931. doi:10.1055/s-2005-864801
- Kamdem Waffo, A. F.; Coombes, P. H.; Crouch, N. R.; Mulholland, D. A.; El Amin, S. M. M.; Smith, P. J. *Phytochemistry* **2007**, *68*, 663. doi:10.1016/j.phytochem.2006.10.011
- Khanye, S. D.; Smith, G. S.; Lategan, C.; Smith, P. J.; Gut, J.; Rosenthal, P. J.; Chibale, K. *J. Inorg. Biochem.* **2010**, *104*, 1079. doi:10.1016/j.jinorgbio.2010.06.005

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/2.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions: (http://www.beilstein-journals.org/bjoc)

The definitive version of this article is the electronic one which can be found at: doi:10.3762/bjoc.7.205